



# BOOK OF PROCEEDINGS



XV International Scientific Agriculture Symposium  
"AgroSym 2024"  
Jahorina, October 10-13, 2024

AgroSym  
2024

AgroSym 2024

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Dear colleagues,

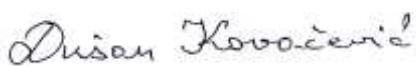
It is with great pleasure that I introduce the *Book of Proceedings* for the 15th International Scientific Agricultural Symposium “AGROSYM 2024.” I trust that you will find it a valuable resource in your work. This year, around 800 contributions have been accepted for inclusion in the *Book of Abstracts*. The themes of AGROSYM 2024 span the full spectrum of agricultural sciences, divided into seven key sessions: 1) Plant Production, 2) Plant Protection and Food Safety, 3) Organic Agriculture, 4) Environmental Protection and Natural Resource Management, 5) Animal Husbandry, 6) Rural Development and Agro-Economy, and 7) Forestry and Agroforestry.

There is growing consensus among scholars and practitioners that technological innovations have the potential to boost agricultural production, enhance supply stability, and reduce the environmental impact of farming. Technology has been especially pivotal in increasing the productivity of annual crops such as maize, rice, soybeans, wheat, and cotton. However, with trees having longer growth cycles, breeding programs for these crops require more time. New plant breeding techniques promise improvements in productivity, but they have sparked an ongoing academic debate regarding their advantages and drawbacks.

While much of the focus in the past has been on the production side of agriculture (e.g., productivity and efficiency), there is an increasing emphasis today on the consumption side and the intermediate stages of the food chain, such as processing and distribution. This shift is driving the transition toward a “farm-to-fork” approach. Globally, consumers are sending clearer signals about their preferences for higher quality, healthier, safer, and more flavorful products. In response, many agri-food companies are exploring innovative ways to gain greater control over production processes and ensure the quality and safety of final products. Furthermore, changes in investment strategies hold the potential to reduce the environmental and social costs of agriculture. It has become increasingly clear to investors that companies that prioritize sustainability and social responsibility tend to yield better long-term returns.

Agri-food systems are at the heart of global discussions surrounding sustainable development and the achievement of the United Nations Sustainable Development Goals (SDGs) by 2030. These systems are deeply connected to numerous global challenges, including climate change, poverty, food insecurity, biodiversity loss, resource depletion, and ecosystem degradation. A key goal of the sustainable agriculture movement is to create farming systems that mitigate or eliminate the environmental harms associated with industrial agriculture. Additionally, it is critical to improve the resilience of food systems to crises, shocks, and pandemics.

I would like to extend my heartfelt thanks to all the authors, reviewers, and colleagues who contributed to the editing of the *Book of Abstracts*. Special thanks are due to our co-organizers and partners for their unwavering collaboration and comprehensive support throughout this endeavor.



Editor in Chief, President of the Scientific Committee of AGROSYM 2024

East Sarajevo, 10 October 2024

Prof. Dušan Kovačević, PhD

## CONTENT

<b>PLANT PRODUCTION .....</b>	<b>31</b>
<b>CROP PRODUCTION: CURRENT STATE AND CHALLENGES, ADVANCES AND PROSPECTS TO BALANCE ENVIRONMENTAL, SOCIAL AND ECONOMIC IMPERATIVES</b>	
Steve QUARRIE .....	32
<b>GENETIC IMPROVEMENT OF DATE PALMS - PROTOPLASTS CULTURE OF TEGGAZA, DEGLET NOUR AND TAQUERBUCHT CULTIVARS</b>	
Djamila YATTA-EL DJOUZI, Said. BOUDEFFEUR, Rosa. SLAMANI, Radia. AYADI, Sana. EZZOUAOUI, Lydia BOUAMRA, Mohamed KHARSSI .....	43
<b>AGRICULTURAL ADAPTATION TO CLIMATE CHANGE IN ALGERIA</b>	
Wassila HANAFI, Nouara BOULFOUL .....	50
<b>FACILITATING HYSSOP INTRODUCTION TO ARMENIA: ASSESSING GLOBAL GENEBAK RESOURCES AND CULTIVATION POTENCIAL</b>	
Alvina AVAGYAN, Laura TADEVOSYAN, Raya BALAYAN, Vladimíra HORČINOVÁ SEDLÁČKOVÁ, Marina HOVHANNISYAN, Margarita HARUTYUNYAN .....	54
<b>A STUDY OF THE EFFECT OF BIOSTIMULANT ON THE YIELD AND QUALITY OF CUCUMBER (CUCUMIS SATIVUS)</b>	
Aleksandra GOVEDARICA-LUČIĆ, Tanja PEROVIĆ, Alma MEMIĆ, Lutvija KARIĆ, Ćerima ZAHIROVIĆ SINANOVIĆ Milica STOJANOVIĆ .....	60
<b>INTERDEPENDENCE OF PRODUCTIVE CHARACTERISTICS AND FORAGE QUALITY OF RED CLOVER</b>	
Borislav PETKOVIĆ, Ilija KOMLJENović, Vojo RADIĆ, Mihajlo MARKOVić, Zoran MALIČEVIĆ .....	65
<b>ECOLOGICAL CHARACTERISTICS OF MACROPHYTES OF THE PROTECTED HABITAT "GROMIŽELJ" (BOSNIA AND HERZEGOVINA)</b>	
Milica LJUBOJEVIC, Nataša MARIC, Sladjana PETRONIC .....	71
<b>ORGANIC PLANT BIOSTIMULANTS AND TOMATO FRUIT QUALITY</b>	
Mirjana JOVOVIĆ, Verica MILOVIĆ, Zoranka MALEŠEVIĆ, Marko PETKOVIĆ .....	77
<b>FLOWERING AND FRUIT SET OF CULTIVARS OF SWEET CHERRY IN AGRO-ECOLOGICAL CONDITIONS OF SARAJEVO</b>	
Mirjana RADOVIĆ, Mihajlo LIZDEK, Mirko KULINA, Dejan ZEJAK, Dragan TOMOVIC, Slavisa MILOVANOVIC .....	84
<b>PROSPECTS OF CORN CULTIVATION AND PRODUCTION IN THE SEMBERIJA REGION IN BOSNIA AND HERZEGOVINA</b>	
Igor ĐURĐIĆ, Nedeljka ELEZ, Angelina PETKOVIĆ, Slađana JANKOVIĆ .....	90
<b>YIELD AND CHARACTERISTICS OF BUNCHES OF RED WINE VARIETIES GROWN IN THE AREA OF TREBINJE (BOSNIA AND HERZEGOVINA)</b>	
Tijana BANJANIN, Snežana BRKLJAĆ, Zorica RANKOVIĆ VASIĆ .....	96

<b>THE YIELD OF POTATOES TESTED IN MULTIPLE ENVIRONMENTS IN REPUBLIKA SRPSKA (BOSNIA AND HERZEGOVINA)</b>	
Branka GOVEDARICA, Tanja MIJATOVIC, Igor ĐURĐIĆ, Vesna MILIĆ, Tanja JAKIŠIĆ .....	103
<b>APPLICATION OF DIFFERENT VARIANTS OF SUPERABSORBENTS AND THEIR IMPACT ON YIELD OF SELECTED POTATO VARIETIES</b>	
Saša LALIĆ, Branka GOVEDARICA, Tanja MIJATOVIC, Igor ĐURĐIĆ, Aleksej LUKIN, Tatiana KRIUKOVA, Vesna MILIĆ .....	109
<b>ENDANGERED AND PROTECTED PLANT SPECIES OF THE RAMSAR SITES OF BOSNIA AND HERZEGOVINA</b>	
Zeljana DOKNIC, Natasa MARIC, Sladjana PETRONIC .....	119
<b>THE EFFECTS OF NaCl AND KCl ON THE GERMINATION AND SEEDLING GROWTH OF <i>AEGILOPS CYLINDRICA</i> HOST SEEDS</b>	
Gergana DESHEVA, Evgeniya VALCHINOVA, Albena PENCHEVA, Bozhidar KYOSEV, Manol DESHEV .....	125
<b>SEEDS QUALITY CHARACTERISTICS OF SOME MEDICINAL PLANTS FROM <i>LAMIACEAE</i> FAMILY USING FLUORESCENCE SPECTROSCOPY</b>	
Katya UZUNDZHALIEVA , Vanya SLAVOVA, Veselina MASHEVA .....	131
<b>COMBINING ABILITY OF MID-EARLY MUTANT MAIZE LINES FOR THE GRAIN YIELD AND LENGTH OF THE EAR</b>	
Mima ILCHOVSKA, Valya BACHIYSKA, Nataliya PETROVSKA, Valentina VALKOVA .....	139
<b>EVALUATION OF LANDRACE AND FOREIGN MAIZE POPULATIONS FOR BIOLOGICAL AND ECONOMIC PROPERTIES</b>	
Yuliana IORDANOVA, Nataliya PETROVSKA .....	146
<b>GRAIN YIELD RESPONSE OF MAIZE, WHEAT AND BARLEY TO NITROGEN RATES AND FERTILIZATION SYSTEMS</b>	
Nedyalka YORDANOVA, Nikolay MINEV, Svetla KOSTADINOVA .....	152
<b>MACRONUTRIENTS ACCUMULATION OF BARLEY DEPENDING ON THE SOIL NITROGEN LEVEL</b>	
Veselin SEVOV, Nikolay MINEV, Svetla KOSTADINOVA .....	158
<b>KNEJA 652 - A LATE HYBRID FROM PORTFOLIO OF THE MAIZE RESEARCH INSTITUTE - KNEZHA FOR 2024</b>	
Valentina VALKOVA .....	164
<b>STUDY OF THE INFLUENCE OF DIFFERENT PERCENTAGE STERILE PLANTS ON THE MAIZE HYBRIDS YIELD</b>	
Valentina VALKOVA Penka VULCHINKOVA, Borislava BACHIYSKA, Zhelyazko VULCHINKOV .....	170
<b>USING OF HETEROSESIS SELECTION IN SESAME (<i>SESAMUM INDICUM L.</i>)</b>	
Stanislav STAMATOV, Veselina MASHEVA Katya UZUNDZHALIEVA .....	175
<b>STABILITY OF BULGARIAN MAIZE HYBRIDS TESTED IN SERBIA</b>	

Stefan VULCHINKOV, Nenad DELIC, Zhelyazko VULCHINKOV, Jovan PAVLOV,  
Natalya PETROVSKA ..... 181

**TRADITIONS OF USING GEORGIAN WHEAT IN GEORGIA**

Levan UJMAJURIDZE, Tsotne SAMADASHVILI, Gulnari CHKHUTIASHVILI ..... 190

**BIOSTIMULANTS ADVERSELY AFFECTED BALSAM ESSENTIAL OIL DURING ITS FIRST YEAR OF ESTABLISHMENT**

Athina TEGOU, Dimitrios BARTZIALIS, Savvas PAPADOPOULOS, Kyriakos D.  
GIANNOULIS, Nicholaos G. DANALATOS, Eleni WOGIATZI..... 194

**BIOSTIMULANTS EFFECT ON LAVENDER YIELD IN ITS SECOND GROWING YEAR**

Kyriakos D. GIANNOULIS, Dimitra-Despoina TOSILIANI, Dimitrios BARTZIALIS,  
Ippolitos GINTSIUDIS, Nicholaos G. DANALATOS ..... 200

**INTERSPECIFIC HYBRIDIZATION BETWEEN *VICIA FABA* L. AND *VICIA NARBONENSIS* L.: PRESENT STATUS AND FUTURE PROSPECTS**

Theano B. LAZARIDOU, Ioannis N. XYNIAS, Demetrios G. ROUPAKIAS ..... 208

**INVESTIGATION OF THE EFFECTS OF DOMESTIC WATER BUFFALO, HUNGARIAN RACKA SHEEP AND HUNGARIAN GREY CATTLE GRAZING ON DIFFERENT WOOD-PASTURES IN HUNGARY**

Attila FŰRÉSZ, Szilárd SZENTES, Dénes SALÁTA, Zsombor WAGENHOFFER,  
Gabriella FINTHA, László SIPOS, Ferenc PAJOR, Károly PENKSZA ..... 213

**BIOLOGICAL DIVERSITY OF *SPINACIA OLERACEA* L. ACCESSIONS COLLECTED IN IRAN USING SOME MORPHOLOGICAL TRAITS**

Naser SABAGHNIA, Mohsen JANMOHAMMADI..... 219

**BIOLOGICAL VARIATION OF SOME GARDEN CRESS (*LEPIDIUM SATIVUM* L.) ACCESSIONS**

Naser SABAGHNIA, Mohsen JANMOHAMMADI..... 223

**GRAPHIC ANALYSIS OF GENOTYPE BY TRAIT INTERACTION IN SAFFLOWER USING BIPILOT METHOD**

Naser SABAGHNIA, Mohsen JANMOHAMMADI..... 227

**MECHANICAL PEAR HARVESTING WITH THE "MULTIBASKET" SYSTEM**

Roberto TOMASONE, Carla CEDROLA, Mauro PAGANO ..... 233

**DEVELOPING A DECISION TREE MOBILE APPLICATION TO CONTROL A SMART GREENHOUSE IN THE JORDAN VALLEY**

Lubna NASIR EDDEEN, Ola SHALBAK, Ansar KHOURI ..... 240

**APPROACHES TO ASSESSING THE ENERGY POTENTIAL OF BIOMASS OBTAINED FROM CROP RESIDUE IN REPUBLIC OF NORTH MACEDONIA**

Igor ILJOVSKI, Zoran DIMOV, Zlatko ARSOV, Ille CANEV ..... 247

**CORRELATION ANALYSIS OF CORN BIO-MORPHOLOGICAL TRAITS IN THE INITIAL GROWTH PHASE UNDER DIFFERENT TEMPERATURE CONDITIONS**

Raisa IVANOVA, Elena LUTCAN, Alla BOROVSKAIA, Nicolai VANICOVICI..... 254

<b>EVALUATING STRAWBERRY CULTIVAR PERFORMANCE AND RUNNER HEALTH ACROSS VARIED CLIMATIC ZONES IN POONCH DISTRICT, AZAD JAMMU AND KASHMIR IN PAKISTAN</b>	
Noosheen ZAHID, Mehdi MAQBOOL, Raheel ANWAR .....	260
<b>ENHANCING APPLE QUALITY ASSESSMENT THROUGH ANOMALY DETECTION WITH AUTOENCODERS</b>	
Adrian IOSIF, Edmond MAICAN, Sorin-Stefan BIRIŞ, Nicolae-Valentin VLĂDUȚ ....	266
<b>EXPLORING INSECTS DETECTION PERFORMANCE IN CORN PEST CONTROL WITH MOBILENET-SSD NEURAL NETWORK AND HIGHER-RESOLUTION INPUT</b>	
Edmond MAICAN, Sanda MAICAN, Sorin-Ştefan BIRIŞ, Adrian IOSIF .....	272
<b>RESEARCH ON DETERMINING THE PERFORMANCE OF A VARIABLE WIDTH PLOUG DESIGNED FOR HIGH-POWER TRACTORS (180-240 HP)</b>	
Sorin-Ştefan BIRIŞ, Constantin LĂCĂTUŞU, Nicolae-Valentin VLĂDUȚ, Neluş-Evelin GHEORGHIȚĂ, Nicoleta UNGUREANU, Edmond MAICAN, Adrian IOSIF .....	279
<b>DIGITAL TECHNOLOGIES IN AGRARIAN SCIENCE AND EDUCATION</b>	
Daniil IAKOVLEV, Alexandr PCHELNICKOV .....	286
<b>INFLUENCE OF STEM CUTTING HEIGHT ON THE NUTRITIONAL VALUE OF CORN SILAGE</b>	
Aleksandar VUKOVIĆ, Dragana LALEVIĆ, Rade STANISAVLJEVIĆ, Saša BARAĆ, Milan BIBERDŽIĆ, Dragoslav ĐOKIĆ .....	292
<b>APPLICATION OF THE SENSOR IN AGRICULTURE: A REVIEW OF RECENT DEVELOPMENTS</b>	
Biljana BOŠKOVIĆ, Irina MARINA, Miloš PAJIĆ, Jonel SUBIĆ .....	299
<b>CONTENT OF MINERAL MATTER IN THE GRAIN OF POPCORN IN A SUSTAINABLE CULTIVATION SYSTEM WITH COVER CROPS</b>	
Biljana SEVIC, Zeljko DOLIJANOVIC, Milena SIMIC, Vesna DRAGICEVIC, Dejan CVIKIC, Jelena STOJILJKOVIC, Ivana ZIVKOVIC .....	304
<b>VARIATION OF SEEDS NUMBER IN PRIMARY SPIKE OF BREAD WHEAT</b>	
Desimir KNEŽEVIĆ, Dragan GRČAK, Milosav GRČAK, Tatjana IVANOVIVIĆ, Mirela MATKOVIĆ STOJŠIN, Danijela KONDIĆ, Adriana RADOSAVAC, Aleksandar PAUNOVIĆ, Veselinka ZEČEVIĆ .....	313
<b>FORECASTING THE IMPACT OF CLIMATE CHANGE ON WATER AVAILABILITY FOR WINTER WHEAT CROP IN THE TOPLICA DISTRICT OF SERBIA</b>	
Dženita KOCA, Gordana MATOVIĆ, Vesna POČUČA, Enika GREGORIĆ, Mirjam VUJADINOVIĆ MANDIĆ .....	321
<b>INFLUENCE OF GENOTYPE AND IRRIGATION ON YIELD COMPONENTS OF SWEET CORN (<i>ZEA MAYS L. VAR. SACCHARATA STURT.</i>)</b>	
Ivan TUPAJIĆ, Marija ČOSIĆ, Aleksa LIPOVAC, Biljana ŠEVIĆ, Jelena STOJILJKOVIĆ, Maja SUDIMAC, Đorđe MORAVČEVIĆ .....	328

<b>MARKER ASSISTED SELECTION FOR <math>\beta</math>-CAROTENE RICH MAIZE: VALIDATION OF MULTIPLEX-PCR ASSAY FOR <i>crtRB1</i> AND <i>lcyE</i> FAVOURABLE ALLELES</b>	
Marija KOSTADINOVIĆ, Danijela RISTIĆ, Milica LUČEV, Dragana IGNJATOVIĆ-MICIĆ, Jelena VANČETOVIĆ.....	334
<b>INFLUENCE OF THE CULTIVAR, FERTILISER, AND IRRIGATION ON ICEBERG LETTUCE MORPHOLOGY - SINGLE FACTOR VS INTERACTION EFFECTS</b>	
Milica STOJANOVIĆ, Zoran DINIĆ, Dragica MILOSAVLJEVIĆ, Aleksandra GOVEDARICA-LUČIĆ, Đorđe MORAVČEVIĆ, Jelena DRAGIŠIĆ MAKSIMOVIĆ, Vuk MAKSIMOVIĆ .....	339
<b>NOVEL BATAVIA LETTUCE CULTIVARS-CHOOSING THE OPTIMAL CULTIVAR FOR FRESH AND PROCESSED PRODUCTS USING WASPAS METHOD</b>	
Milica STOJANOVIĆ, Dragica MILOSAVLJEVIĆ, Jelena DRAGIŠIĆ MAKSIMOVIĆ, Vuk MAKSIMOVIĆ, Aleksandra GOVEDARICA-LUČIĆ, Radomir BODIROGA .....	346
<b>YIELD RESPONSE OF MAIZE TO IRRIGATION AND PLANTING DENSITY IN THE VOJVODINA ENVIRONMENT IN SERBIA</b>	
Miodrag TOLIMIR, Branka KRESOVIĆ, Zorica SREDOJEVIĆ, Katarina GAJIĆ, Milan BRANKOV, Marijana DUGALIĆ, Boško GAJIĆ .....	353
<b>STRESS RESISTANCE INDICATORS AS THE TOOL FOR SELECTING DROUGHT-TOLERANT WHEAT GENOTYPES</b>	
Mirela MATKOVIĆ STOJŠIN, Veselinka ŽEČEVIĆ, Dušan UROŠEVIĆ, Dragan BOŽOVIĆ, Jasmina BAČIĆ, Kamenko BRATKOVIĆ, Desimir KNEŽEVIĆ.....	361
<b>SOCIAL NETWORKING IN STRESS-EXPOSED PLANTS</b>	
Mirjana VUKOSAVLJEV, Brankica KARTALOVIĆ, Nevena STEVANović, Maša BUĐEN, Nikola STANKOVIĆ, Boris BRKIĆ, Nataša LJUBIČIĆ .....	367
<b>THE NORMALIZED DIFFERENCE RED EDGE INDEX (NDRE) IN GRAIN YIELD AND BIOMASS ESTIMATION IN MAIZE (<i>Zea mays L.</i>)</b>	
Nataša LJUBIČIĆ, Vera POPOVIĆ, Marko KOSTIĆ, Mirjana VUKOSAVLJEV, Maša BUĐEN, Nikola STANKOVIĆ, Nevena STEVANović.....	373
<b>MONITORING STRESS IN ARUGULA (<i>Eruca sativa</i>) USING A PORTABLE MULTISPECTRAL DEVICE</b>	
Nevena STEVANović, Nataša LJUBIČIĆ, Nikola STANKOVIĆ, Maša BUĐEN, Brankica KARTALOVIĆ, Mirjana VUKOSAVLJEV .....	379
<b>EFFECTS OF NITROGEN FERTILIZATION ON THE YIELD OF TRADITIONAL VARIETIES AND LANDRACES OF WINTER WHEAT</b>	
Sanja MIKIĆ, Milan MIROSAVLJEVIĆ, Ljiljana BRBAKLIĆ, Vladimir AĆIN, Jelena VISKOVIĆ, Goran JAĆIMOVIĆ .....	384
<b>RESULTS OF TESTING THE WORKING QUALITY OF DIFFERENT TECHNOLOGICAL-TECHNICAL SYSTEMS FOR PRECISION SOWING OF MAIZE</b>	
Saša BARAĆ, Milan BIBERDŽIĆ, Aleksandar ĐIKIĆ, Aleksandar VUKOVIĆ, Dragana LALEVIĆ .....	392

<b>SERBIAN TOMATO PRODUCTION GROSS MARGIN ANALYSIS: ECONOMIC VIABILITY AND SUSTAINABILITY OF OPEN-FIELD VS. GREENHOUSE</b>	
Sladan STANKOVIC, Vedran TOMIC, Nikola LJILJANIĆ, Dusan NIKOLIĆ, Slobodan KRNJAJIĆ, Elmira SALJNIKOV, Vesela RADOVIĆ .....	398
<b>SERBIA AGRICULTURE SECTOR TRANSFORMATION DUE TO GLOBAL, CLIMATE AND EU CHALLENGES</b>	
Sladan STANKOVIC, Rade JOVANOVIĆ, Snezana JANKOVIĆ, Divna SIMIĆ, Dusan NIKOLIĆ, Aleksandra IVETIĆ, Jelena IVANOVIC, Stefana UTVIĆ .....	403
<b>BIOCHEMICAL AND MOLECULAR CHARACTERIZATION OF MAIZE INBRED LINES</b>	
Snezana MLADENOVIĆ DRINIĆ, Jelena VUKADINOVIĆ, Danijela RISTIĆ, Natalija KRAVIĆ, Violeta ANĐELOKOVIĆ, Nikola GRČIĆ, Ana NIKOLIĆ.....	410
<b>ISOLATION AND CHARACTERIZATION OF PGP BACTERIA FROM <i>PLANTAGO LANCEOLATA</i> RHIZOSPHERE SOIL</b>	
Timea HAJNAL JAFARI, Vladimira ŽUNIĆ, Dragana STAMENOV, Simonida ĐURIĆ .....	417
<b>THE INFLUENCE OF AUXIN OF BACTERIAL ORIGIN ON THE SEED QUALITY PARAMETERS OF SOME FIELD CROPS</b>	
Violeta MANDIĆ, Snezana ĐORĐEVIĆ, Nikola ĐORĐEVIĆ, Vesna KRNJAJA, Milica BALOŠ .....	424
<b>THE INFLUENCE OF HEAVY METALS ON THE GROWTH OF THE WHEAT STEM</b>	
Violeta MICKOVSKI STEFANOVIĆ, Predrag BRKOVIĆ, Vladimir FILIPOVIĆ, Aleksa ĐUKIĆ, Nenad ŽIVKOVIĆ, Dragana STEVANOVIĆ .....	432
<b>FORAGE YIELD AND WEEDNESS OF THE INTERCROPPED SPRING VETCH AND OATS</b>	
Vladeta STEVOVIĆ, Dalibor TOMIĆ, Bojana LAPČEVIĆ, Miloš MARJANOVIĆ, Nenad PAVLOVIĆ, Milomirka MADIĆ .....	437
<b>EFFECT OF DIFFERENT MICROBIAL FERTILIZERS ON THE MORPHOLOGICAL AND PRODUCTIVE CHARACTERISTICS OF MAIZE</b>	
Željko DOLIJANOVIĆ, Vesna DRAGIČEVIĆ, Snezana ĐORĐEVIĆ, Snezana OLJAČA, Srđan SEREMEŠIĆ .....	443
<b>EFFECT OF AMMONIUM THIOSULFATE NITROGEN FERTILIZER ON QUALITATIVE PARAMETERS OF AROMATIC PLANTS</b>	
Ivana MEZEYOVÁ, Ivana KOLLÁROVÁ, Ján MEZEY, Miroslav ŠLOSÁR, Mária BREZÁNIOVÁ .....	449
<b>EFFECTIVENESS OF BIOSTIMULANTS ON THE QUANTITATIVE AND QUALITATIVE PARAMETERS OF BLUEBERRY (<i>VACCINUM CORYMBOSUM</i>)</b>	
Ján MEZEY, Petra ŠEVČÍKOVÁ, Andrej SELNEKOVIĆ, Ivana MEZEYOVÁ .....	456
<b>EFFECT OF CULTIVAR AND PROCESSING METHOD ON QUALITY OF CHIPS MADE FROM TURNIP, RUTABAGA AND RADISH</b>	
Miroslav ŠLOSÁR, Júlia FABIANOVÁ, Ivana MEZEYOVÁ, Ján MEZEY, Ivana TIRDILOVÁ .....	466

<b>CORRELATION ANALYSIS OF CORM DENSITY AND FERTILIZATION TREATMENTS ON SAFFRON GROWTH AND YIELD IN SEMI-ARID CLIMATE</b>	
Youssef ABOU OBEID, Layla NAIM, Nidal SHABAN, Grigor ZEHIROV, Fadi Sami KARAM .....	473
<b>IMPACT OF CLIMATE CHANGE ON SUNFLOWER PRODUCTION</b>	
Vojin CVIJANOVIĆ, Marija BAJAGIĆ, Mladen PETROVIĆ, Gorica CVIJANOVIĆ, Bojan DIMITRIJEVIĆ .....	479
<b>ENSEMBLE EVALUATION AND MEMBER SELECTION OF REGIONAL CLIMATE MODELS FOR CROP MODEL IMPACT ASSESSMENT</b>	
Melpomeni NIKOU, Aristeidis K GEORGULIAS, Vassilis G ASCHONITIS, Dimitris M PAPAMICHAIL, Theodoros MAVROMATIS .....	486
<b>THE EFFECT OF CHANGES IN ENVIRONMENTAL CONDITIONS ON THE GROWTH AND YIELD OF BELL PEPPER</b>	
Zdenka GIREK, Suzana PAVLOVIĆ, Jelena DAMNJANOVIĆ, Dejan CVIKIĆ, Milan UGRINOVICIĆ .....	493
<b>PLANT PROTECTION AND FOOD SAFETY .....</b>	<b>499</b>
<b>MORPHOLOGICAL AND MOLECULAR IDENTIFICATION OF ECTOPARASITIC NEMATODES ASSOCIATED WITH THE GENUS XIPHINEMA SPECIES IN KOSOVO VINEYARDS</b>	
Gazmend GJINOVCI, Adnand RAMADHI, Shpend SHAHINI, Betim BRESILLA, Agron BUNJAKU, Bekri XHEMALI .....	500
<b>PHYTOTOXIC EFFECTS OF THYMUS VULGARIS AND CITRUS SINENSIS ESSENTIAL OILS ON SEED GERMINATION OF SEVEN CEREAL VARIETIES</b>	
Fatma Zohra HAMDANI, Namira MHAMEDI BOUZINA, Imane KRAOUCHE.....	506
<b>MICROBIOLOGY OF WATER IN PRIMARY FOOD PRODUCTION</b>	
Bojan GOLIĆ, Dragan KNEŽEVIĆ, Biljana PEĆANAC .....	514
<b>HARMFUL ENTOMOFAUNA ON STRAWBERRY IN BIJELJINA AREA (BOSNIA AND HERZEGOVINA)</b>	
Dejana STANIĆ, Milanka MAŠANOVIC .....	520
<b>EFFECT OF CEREAL-LEGUME INTERCROPPING AND MARIGOLD TAGETES SP. ON SOIL PLANT PARASITIC NEMATODES</b>	
Georgi DIMITROV, Lilyana KOLEVA .....	531
<b>THE EFFECT OF ESSENTIAL OILS ON MYCELIAL GROWTH AND SPORULATION OF BOTRYTIS CINerea</b>	
Slavko GRGIĆ, Tamara SIBER, Dunja ĆOSIĆ, Karolina VRANDEČIĆ, Elena PETROVIĆ, Jasenka ĆOSIĆ .....	538
<b>IN VIVO EVALUATION OF ANTIFUNGAL ACTIVITY OF NICOTINAMIDE COMPOUNDS AGAINST BOTRYTIS CINerea IN TOMATOES</b>	
Tamara SIBER, Elena PETROVIĆ, Jasenka ĆOSIĆ, Karolina VRANDEČIĆ .....	544
<b>THE LIPOLYSIS LEVEL OF SOME GREEK ARTISANAL CHEESES: A REVIEW OF LAST YEAR'S DATA</b>	

Aikaterini GEORGALA .....	550
<b>INFLUENCE OF FEED FORM ON YELLOW MEALWORM ADULT PRODUCTIVITY: PRELIMINARY RESULTS</b>	
Fjolla AVDYLAJ, Flutura LAMAJ, Ferdinando BALDACCHINO .....	556
<b>NATURAL ENEMIES OF THE FALL ARMY WORM <i>SPODOPTERA FRUGIPERDA</i> (LEPIDOPTERA: NOCTUIDAE) IN JORDAN AND PALESTINE</b>	
Ahmad KATBEH-BADER, Ibrahim AL-JBOORY, Thaer YASEEN .....	561
<b>EVALUATION OF <i>PSEUDOMONAS</i> FOR THE MANAGEMENT OF TOMATO DISEASE AND PESTS</b>	
Redouan QESSAOUI, Salahddine CHAFIKI, Abdelmalk MAHROUG, Abdelhadi AJERRA, Abdelghani TAHIRI, Naima AIT AABD, Mohamed ALOUANI, Bouchra CHEBLI, Rachid BOUHARROUD .....	567
<b>PHOSPHATE SOLUBILIZATION BY THE FLUORESCENTS <i>PSEUDOMONAS</i></b>	
Redouan QESSAOUI, Salahddine CHAFIKI, Naima CHABBBI, Abdelmalek MAHROUG, Abdelghani TAHIRI, Naima AIT AABD, Mohamed ALOUANI, Rachid BOUHARROUD .....	573
<b>ANTIOXIDANT ACTIVITY OF EXTRACTS FROM DIFFERENT COMPONENTS OF THE INFRUCTESCENCES OF TWO CHESTNUT CULTIVARS</b>	
Elsa RAMALHOSA, Filipe LEMA, Maria Halana FLORINDO DA SILVA, Filomena BARREIRO, Arantzazu SANTAMARIA-ECHART, Tatiana SCHREINER, Maria do Céu FIDALGO, Cristina OLIVEIRA, Paula BAPTISTA .....	579
<b>2022 FARMERS' WILLINGNESS TO REDUCE THE USE OF HERBICIDES IN MANAGING WEEDS IN RICE CROPS</b>	
Maria DE FÁTIMA OLIVEIRA, Isabel CALHA, Pedro REIS .....	585
<b>EXPLORING INSECTS DETECTION PERFORMANCE IN CORN PEST CONTROL WITH MOBILENET-SSD NEURAL NETWORK AND HIGHER-RESOLUTION INPUT</b>	
Edmond MAICAN, Sanda MAICAN, Sorin-Ştefan BİRİŞ, Adrian IOSIF .....	593
<b>IMPROVING ENERGY EFFICIENCY IN ROMANIAN AGRICULTURE: MECHANICAL ENGINEERING'S CHALLENGES AND CONTRIBUTIONS</b>	
Neluş-Evelin GHEORGHIȚĂ, Sorin-Ştefan BİRİŞ, Nicoleta UNGUREANU, Mariana IONESCU .....	600
<b>PNEUMATIC CONVEYING SYSTEMS IN THE AGRI-FOOD INDUSTRY – STATE OF THE ART</b>	
Raluca-Adriana ZOTA, Sorin-Ştefan BİRİŞ, Neluş-Evelin GHEORGHIȚĂ .....	606
<b>PLECTOSPHEARELLA CUCUMERINA – NEW AND EMERGING PATHOGEN OF LETTUCE IN BALKAN REGION</b>	
Milica MIHAJLOVIĆ, Jovana HRUSTIĆ, Ana VUČUROVIĆ, Maja ŽIVANOVIĆ, Brankica PEŠIĆ .....	614
<b>EFFICACY OF PYRAFLUFEN-ETHYL IN THE CONTROL OF SUCKERS IN RASPBERRY, APPLE AND PLUM ORCHARDS IN SERBIA</b>	

Ana ANĐELKOVIĆ, Dragana MARISAVLJEVIĆ, Stefan KOVAČEVIĆ, Danijela ŠIKULJAK .....	620
<b>EFFECTS OF THE INSECTICIDE LAMBDA-CYHALOTHRIN IN CONTROL OF APHIS POMI DE GEER IN APPLE ORCHARDS</b>	
Slavica VUKOVIĆ, Dragana ŠUNJKA, Antonije ŽUNIĆ, Miloš PETROVIĆ, Nikola LAĆARAC, Dragana BOŠKOVIĆ, Aleksandra ŠUŠNjar, Jelena EĆIMOVIĆ .....	625
<b>PARAQUAT AND OTHER DESSICANTS – INFLUENCE ON PLANTS AND OTHER ORGANISMS</b>	
Bogdan NIKOLIĆ, Sanja ĐUROVIĆ, Boris PISINOV, Vladan JOVANOVIĆ, Tijana DUDIĆ, Miloš DUGALIĆ .....	629
<b>THE ROLE OF VOLATILE ORGANIC COMPOUNDS IN PLANT-SOIL COMMUNICATION</b>	
Brankica KARTALOVIĆ Mirjana VUKOSAVLJEV, Nevena STEVANOVIĆ, Boris BRKIĆ, Nikola STANKOVIĆ, Maša BUĐEN, Nataša LJUBIČIĆ .....	639
<b>APPLICATION OF MATHEMATICAL DECONVOLUTION ON FTIR SPECTRA FOR DETERMINATION OF PROTEIN STRUCTURE IN TOMATO (<i>Solanum lycopersicum L.</i>) TREATED WITH <i>TRICHODERMA SPP.</i></b>	
Daniela DJIKANOVIĆ, Igor VUKELIĆ, Danka RADIĆ, Ilinka PEĆINAR, Steva LEVIĆ, Dejana PANKOVIĆ, Ksenija RADOTIĆ .....	644
<b>CONTACT TOXICITY OF THE <i>PINUS SIBIRICA</i> ESSENTIAL OIL AGAINST <i>DROSOPHILA SUZKII</i></b>	
Dragana BOŠKOVIĆ, Nuray BASER, Slavica VUKOVIĆ, Sanja LAZIĆ, Aleksandra ŠUŠNjar, Dušan ČULUM, Antonije ŽUNIĆ, Jelena EĆIMOVIĆ, Dragana ŠUNJKA	650
<b>PRELIMINARY STUDY OF ESSENTIAL OILS EFFECT ON <i>BOTRYTIS CINerea</i></b>	
Jelena EĆIMOVIĆ, Dragana BOŠKOVIĆ, Slavica VUKOVIĆ, Sanja LAZIĆ, Aleksandra ŠUŠNjar, Antonije ŽUNIĆ, Dragana ŠUNJKA .....	656
<b>MYCOPOPULATION ON RASPBERRIES IN SERBIA</b>	
Tanja VASIĆ, Sanja ŽIVKOVIĆ, Darko JEVREMOVIĆ, Aleksandra LEPOSAVIĆ, Stefan KOVAČEVIĆ, Jordan MARKOVIĆ, Aleksandra BULAJIĆ .....	661
<b>PHYTOPATOGENIC FUNGI OF ALFALFA IN SERBIA</b>	
Tanja VASIĆ, Sanja ŽIVKOVIĆ, Jordan MARKOVIĆ, Sonja FILIPOVIĆ, Debasis MITRA .....	667
<b>INFLUENCES OF DIFFERENT DRYING METHODS ON THE CHEMICAL COMPOSITION OF BURLEY TOBACCO</b>	
Vesna RADOJIČIĆ, Tijana UROŠEVIĆ .....	673
<b>CHEMICAL TRANSFORMATIONS IN FLUE-CURED TOBACCO DEPENDING ON THE DRYING METHOD: A REVIEW</b>	
Vesna RADOJIČIĆ, Marija SRBINOSKA, Lazar PEJIĆ, Tijana UROŠEVIĆ .....	678
<b>MICROBIOLOGICAL QUALITY OF HOMOLJA WHITE CHEESE IN BRINE</b>	
Višnja SIKIMIĆ, Slavica ČABRILO, Aleksandra MILUTINOVIĆ, Nada JELIĆ .....	684

<b>CORRELATION BETWEEN TOTAL POLYPHENOLS CONTENT AND TOTAL FLAVONOIDS CONTENT IN RED GRAPEFRUIT WITH ANTIOXIDANT ACTIVITY</b>	
Vojkan MILJKOVIĆ, Ivana GAJIĆ, Radomir LJUPKOVIĆ, Jelena MRMOŠANIN, Milena MILJKOVIĆ .....	691
<b>PEARSON'S CORRELATION COEFFICIENT FOR TOTAL POLYPHENOLS AND TOTAL FLAVONOIDS WITH ANTIOXIDANT ACTIVITY IN TOMATO JUICE</b>	
Vojkan MILJKOVIĆ, Ivana GAJIĆ, Jelena MRMOŠANIN, Milena MILJKOVIĆ .....	694
<b>CORDYCEPS SINENSIS MUSHROOMS, THEIR ANTIOXIDANT, ANTIMICROBIAL ACTIVITY AND COMPLEXATION WITH SILVER IONS</b>	
Anna UHRINOVÁ, Lucia UNGVARSKÁ MALUČKÁ.....	698
<b>SURVEILLANCE OF RHIZOCTONIA SOLANI STRAIN PATHOGENICITY ON DIFFERENT CROP SPECIES AND BIOCONTROL POTENTIAL OF CANDIDATE AGENTS</b>	
Abdelhak RHOUMA, Lobna HAJJI-HEDFI, Omaima BARGOUGUI, Amira KHLIF, Samar DALI .....	704
<b>EFFECTIVENESS OF PSEUDOMONAS SP. CULTURE FILTRATE IN THE MANAGEMENT OF ALTERNARIA SOLANI CAUSING TOMATO EARLY BLIGHT</b>	
Abdelhak RHOUMA, Lobna HAJJI-HEDFI, Omaima BARGOUGUI, Amira KHLIF, Samar DALI .....	712
<b>A REVIEW OF UTILIZING ENTOMOPATHOGENIC NEMATODES FOR CONTROLLING STORAGE PESTS</b>	
Alperen Kaan BÜTÜNER, İsmail Alper SUSURLUK.....	720
<b>OLFACTORY RESPONSE OF AN IMPORTANT TURF PEST, DORCADION PSEUDOPREISSI BREUNING (COL.: CERAMBYCIDAE) TO DIFFERENT ESSENTIAL OILS</b>	
Esin SERT, Nimet Sema GENÇER .....	726
<b>ASSESSMENT OF NON-INTENTIONALLY ADDED SUBSTANCES MIGRATION FROM PLASTIC PACKAGING TO FOOD</b>	
Arca AKBAŞ, Özlem KIZILIRMAK ESMER .....	732
<b>EXTENDING THE POSTHARVEST QUALITY OF MINIMALLY PROCESSED (SLICED) PEACHES BY POSTHARVEST ASCORBIC ACID IMMERSION AND CHITOSAN COATING TREATMENTS</b>	
Sevil UNAL, Ferhan K. SABIR .....	739
<b>METHYL JASMONATE MAINTAINS THE POSTHARVEST PHYSICAL AND BIOCHEMICAL QUALITY PROPERTIES OF 'ALPHONSE LAVELLÉE' (<i>Vitis vinifera</i> L.) TABLE GRAPES DURING COLD STORAGE</b>	
Sevil UNAL, Ferhan SABIR, Zahidenur TURAN, Ali SABIR .....	746
<b>DETERMINATION METHOD OF HEXAKANAZOL AND TEBUCUNAZOL RESIDUES BY CAPILAR ELECTROPHORESIS</b>	
Ülkü Dilek UYSAL ÖZDEMİR, Tufan GÜRAY .....	753

<b>DETERMINATION METHOD OF HEXACONAZOLE AND TEBUCUNAZOL RESIDUES BY CAPILLARY ELECTROPHORESIS</b>	
Tufan GÜRAY, Ulku Dilek UYSAL .....	759
<b>SUPPRESSION OF OSTRINIA NUBILALIS IN CORN WITH UAV – SPRAY APPLICATION</b>	
Zoran STOJANOVIĆ, Momir ALVIROVIĆ, Gordana ĐORĐEVIĆ, Mladen PETROVIĆ, Vojin CVIJANOVIC .....	766
<b>APPLICATION OF BACILLUS spp. IN PLANT PROTECTION AND GROWTH PROMOTION OF CEREALS</b>	
Marina JOVKOVIĆ, Magdalena KNEŽEVIĆ, Marina DERVIŠEVIĆ, Jelena MAKSIMOVIĆ, Galina JEVĐENOVIC, Mira MILINKOVIĆ, Aneta BUNTIĆ .....	773
<b>ORGANIC AGRICULTURE.....</b>	<b>779</b>
<b>VARIATION OF ORGANIC CARBON CONCENTRATION IN BUCKWHEAT GROAT WASTE EXTRACTS</b>	
Odetta POCIENE, Rasa SLINKSIENE .....	780
<b>EFFECTS OF HARVESTING TIME ON ESSENTIAL OIL YIELD AND COMPONENTS OF ORGANICALLY GROWN THYME (<i>Thymus vulgaris L.</i>)</b>	
Mehmet ARSLAN, Sancar BULUT.....	787
<b>USE OF ESSENTIAL OIL FOR WEED CONTROL IN ORGANIC FARMING</b>	
Mehmet ARSLAN, Sancar BULUT.....	793
<b>ENVIRONMENT PROTECTION AND NATURAL RESOURCES MANAGEMENT.....</b>	<b>798</b>
<b>EVALUATION OF GROUNDWATER QUALITY OF THE TIRANA (ALBANIA) WASTE TREATMENT AREA</b>	
Ornela SHOSHI, Greta MALAJ.....	799
<b>SOLID-STATE FERMENTATION CELLULASES PRODUCTION ON WHEAT BRAN</b>	
Nabila BELHAMICHE, Souheila OUANAS, Francis DUCHIRON, Said BENALLAOUA .....	808
<b>INVASIVE SPECIES IN THE VINEYARDS OF BOSNIA AND HERZEGOVINA</b>	
Mate BOBAN, Helena BREKALO, Sandra MEDIC, Antonela MUSA, Danijela PETROVIĆ.....	812
<b>STRUCTURE OF AGRICULTURAL LAND LOSSES IN FORMER YUGOSLAVIA COUNTRIES OVER THE PERIOD 1990-2018</b>	
Branislav DRAŠKOVIĆ, Miroslav LALOVIĆ, Milan GLIŠIĆ, Ljiljana TANASIĆ, Darko MARKOVIĆ .....	818
<b>THE LIFE CYCLE OF THAUMETOPOEA PITYOCAMPA (DENIS &amp; SCHIFFERMÜLLER, 1775) ON THE PINE TREES OF THE CITY OF MOSTAR AND THE HEALTH RISK</b>	
Alma ČOLAK, Denisa ŽUJO ZEKIĆ Emina ADEMOVIĆ, Alisa HADŽIABULIĆ, Elma TEMIM .....	824

<b>ALLERGENIC WOODY PLANT SPECIES IN THE COURTYARDS OF SCHOOLS IN THE CITY OF MOSTAR IN BOSNIA AND HERZEGOVINA</b>	
Danijela PETROVIĆ, Helena BREKALO, Sandra MEDIĆ, Antonela MUSA, Toni GALIĆ, Mate BOBAN .....	831
<b>EFFECT OF FERTILIZATION AND TYPE OF GRASS-LEGUMINOUS MIXTURE ON GRASSLAND COVER ON TECHNOGENIC SOIL</b>	
Nenad MALIĆ, Una MATKO, Ivana TOŠIĆ .....	837
<b>SANITARY WASTEWATER FROM CAMPS AT HIGHWAY CONSTRUCTION SITES</b>	
Zoranka MALEŠEVIĆ, Mirjana JOVOVIĆ, Selena ĆEVRIZ .....	846
<b>EVALUATION OF SERUM CYSTATIN C, SYMMETRIC DIMETHYLARGININE (SDMA), CREATININE AND UREA IN CLINICALLY HEALTHY GERIATRIC HORSES</b>	
Adelina KARASTOEVA, Sasho SABEV .....	852
<b>ULTRAFILTRATION SEPARATION AND APPLICATION OF WASTE POLYPHENOLS FROM OIL-BEARING ROSES</b>	
Milena MITEVA, Yana KOLEVA, Viktoria TRIFONOVA, Ana DOBREVA .....	859
<b>CLIMATE NEUTRALITY AND THE ROLE OF EMISSIONS TRADING IN THE EUROPEAN UNION</b>	
Monika SABEVA, Diana KOPEVA .....	865
<b>EVALUATION OF PHYTOREMEDIATION POTENTIAL OF <i>SIDA HERMAPHRODITA</i> GROWN IN CONTAMINATED SOILS WITH HEAVY METALS</b>	
Violina ANGELOVA, Vera KOLEVA .....	871
<b>CHEMICAL PROPERTIES AND POTENTIALLY TOXIC ELEMENTS IN ANTHROPOGENIC SOILS ON FLYSCH DEPOSITS IN OLIVE GROVES OF KAŠTELA BAY, CROATIA</b>	
Aleksandra BENSA, Katarina MATAN, Nikolina JURKOVIĆ BALOG, Aleksandra PERČIN .....	877
<b>SOIL PROPERTIES AND SOIL FERTILITY EVALUATION OF BUKOVICA REGION, CROATIA</b>	
Danijela JUNGIĆ, Jurica ŠKRLEC, Stjepan HUSNJAK, Aleksandra BENSA .....	884
<b>STUDY OF MICROPLASTICS IN SOIL AMENDEMENTS</b>	
Ioanna CHOMATA, Chrysi A. P APADIMITRIOU .....	890
<b>PRODUCTION PRACTICES OF ECONOMICALLY IMPORTANT CROPS OF GREECE AS A POTENTIAL SOURCE OF MICROPLASTICS IN THE SOIL ECOSYSTEM</b>	
Christos ARAMPATZIS, Ekavi ISARI, Petros KOKKINOS, Ioannis KALAVROUZIOTIS .....	895
<b>ENHANCING SOIL BULK DENSITY PREDICTION: A COMPARATIVE PRELIMINARY STUDY OF ML METHODS AND PEDOTRANSFER FUNCTIONS</b>	
Melpomeni NIKOU, Panagiotis TZIACHRIS, Eirini METAXA .....	901

<b>PRODUCTION TECHNOLOGY OF AN INNOVATIVE SOIL FERTILITY IMPROVEMENT MEANS</b>	
Aleksandrs ADAMOVICS, Janis MILLERS .....	908
<b>GREEN LOGISTICS PRINCIPLES AND PRACTICES IN LOGISTICS COMPANIES IN LATVIA</b>	
Andra ZVIRBULE, Gunta GRINBERGA-ZALITE, Anita AUZINA.....	914
<b>TOWARDS SUSTAINABLE CONSUMPTION: INSIGHTS FROM LATVIAN WASTE MANAGEMENT PRACTICES</b>	
Gunta GRINBERGA-ZALITE, Andra ZVIRBULE, Anita AUZINA.....	921
<b>FORECAST POST-2022 BARLEY PRODUCTION IN LIBYA USING TIME SERIES ANALYSIS TO ACHIEVING SELF -SUFFICIENCY</b>	
Jamal Ali Mohamed EHDADAN.....	928
<b>THE ROLE OF WHEAT IN LIBYA AND THE IMPACT ON FOOD SECURITY: CHALLENGES AND OPPORTUNITIES</b>	
Jamal Ali Mohamed EHDADAN.....	933
<b>A SYSTEMATIC REVIEW OF THE IMPACT OF CLIMATE CHANGE ON AGRICULTURE AND FUTURE ADAPTATION STRATEGIES FOR NORTH MACEDONIA</b>	
Elizabeta DIMITRIESKA STOJKOVIKJ, Radmila CRCEVA NIKOLOVSKA, Risto UZUNOV, Sandra MOJSOVA, Aleksandra ANGELESKA, Ljupco ANGELOVSKI....	939
<b>TRANSLOCATION OF ENDANGERED SPECIES <i>SERRATULA LYCOPIFOLIA</i> (VILL.) A. KERN. IN THE REPUBLIC OF MOLDOVA</b>	
Olga IONITA, Elena TOFAN-DOROFEEV, Veaceslav GHENDOV .....	948
<b>LEVERAGING BLOCKCHAIN TECHNOLOGY TO ENHANCE SUSTAINABILITY AND TRACEABILITY IN GLOBAL AGRI-FOOD VALUE CHAINS</b>	
Manal BELATIK, Mustapha OUATMANE .....	954
<b>CLIMATE CHANGE IMPACT ON VEGETABLE CROPS AND POTENTIAL FOR ADAPTATION</b>	
Aanuoluwapo AYOBAMI OWOLABI, Funmilayo MARY OLOYEDE .....	960
<b>GROUNDWATER QUALITY FOR IRRIGATION AND DOMESTIC USES OF AIR MOUNTAINS IN SEMI ARID REGION (AGADEZ, NIGER)</b>	
Alhassane ILLIAS, Maman Sani ABDOU BABAYE, Souleymane ISSA MALAN S....	974
<b>BIOPHYSICAL CHARACTERIZATION OF THE IRRIGATION ALONG THE TELWA VALLEY, IN THE DEPARTEMENT OF TCHIROZERINE, AGADEZ</b>	
Mahamadou MOUSSA DIT KALAMOU, Ama Souleymane ABOUBACAR, Maharazu A. YUSUF .....	980
<b>Characteristics of the Study Area.....</b>	<b>982</b>
<b>Irrigated areas along the stream in the three municipalities .....</b>	<b>982</b>
<b>OPTIMIZING PHOTOSYNTHETIC MICROBIAL FUEL CELLS: EFFECTS OF ELECTRIC STIMULATION ON BIOFILM GROWTH</b>	

Adam STAROWICZ, Paulina RUSANOWSKA, Marcin ZIELIŃSKI .....	987
<b>THE INFLUENCE OF AIR TEMPERATURE TRENDS ON CHERRY PLUM (<i>PRUNUS CERASIFERA EHRH</i>) IN THE BLUE-GREEN INFRASTRUCTURE</b>	
Mirjana OCOKOLJIĆ, Đurđa PETROV, Nevenka GALEČIĆ, Dejan SKOČAJIĆ, Dragan VUJIČIĆ, Isidora SIMOVIĆ .....	1000
<b>GREEN INFRASTRUCTURE: IMPLEMENTATION OF WINTERSWEET IN LANDSCAPE DESIGN AND ENHANCEMENT OF ECOSYSTEM SERVICES</b>	
Mirjana OCOKOLJIĆ, Đurđa PETROV, Nevenka GALEČIĆ, Dejan SKOČAJIĆ, Dragan VUJIČIĆ, Isidora SIMOVIĆ .....	1007
<b>TILLAGE SYSTEMS - ADVANTAGES AND DISADVANTAGES IN TERMS OF NUTRIENT CONTENT, YIELD AND ECONOMIC PROFITABILITY</b>	
Dragana LALEVIĆ, Milan BIBERDŽIĆ, Aleksandar VUKOVIĆ, Saša BARAĆ, Lidija MILENKOVIĆ, Zoran S. ILIĆ, Olivera ŠUŠA .....	1014
<b>SOIL WATER RETENTION IN PANNONIAN CHERNOZEM</b>	
Vladimir ĆIRIĆ, Dragana MARINKOVIĆ, Dragan RADOVANOVIC .....	1021
<b>ASSESSING THE INFLUENCE OF NITROGEN PRESSURE ON NITROGEN COMPOUND LEVELS AND COMPOSITION IN SELECTED SURFACE WATER BODIES</b>	
Marija PEROVIĆ, Vesna OBRADOVIĆ .....	1027
<b>NITROGEN PRESSURE IMPACT ON GROUNDWATER NITROGEN COMPOUND LEVELS</b>	
Marija PEROVIĆ, Vesna OBRADOVIĆ .....	1033
<b>BIO-CONTROL OF THE FIVE MOST INVASIVE WEEDS AS A NARRATIVE FOR THE SUSTAINABLE GOVERNANCE OF BALKAN NATURAL RESOURCES</b>	
Renata GAGIĆ-SERDAR, Miroslava MARKOVIĆ, Bojan KONATAR, Suzana MITROVIĆ, Danilo FURTULA, Ljubinko RAKONJAC .....	1039
<b>RESTORATION AND CONSERVATION OF THE PROTECTED TREE <i>QUERCUS ROBUR</i> L. AT THE SITE "JOŽIĆA KOLIBA", SERBIA</b>	
Suzana MITROVIĆ, Milorad VESELINOVIC, Zlatan RADULOVIC, Nevena ČULE, Snežana STAJIĆ, Miroslava MARKOVIĆ, Marija MILOSAVLJEVIĆ .....	1047
<b>EVALUATING CORROSION AND INCrustATION RISKS IN SELECTED WELLS IN DANUBE ALLUVIUM</b>	
Vesna OBRADOVIĆ, Marija PEROVIĆ, Predrag PAJIĆ .....	1054
<b>SHEEP WOOL BRIQUETTES</b>	
Dominika GORNIK BUČAR, Janij KAVČIĆ, Bojan GOSPODARIĆ .....	1059
<b>SPATIOTEMPORAL VARIATION OF SNOWFALL AND SNOW COVER IN URBAN CENTERS OF THE WESTERN BLACK SEA REGION, TURKEY</b>	
Hüseyin SENSOY, İlyas BOLAT .....	1065
<b>THE EFFECT OF THE COMBINED USAGE OF HYDROFOBIC AND HYDROPHILIC AGENTS ON THE STALING OF BREAD</b>	
Sultan ARSLAN-TONTUL .....	1071

<b>LESS CLEAN WATER FOR AGRICULTURE DUE TO MILITARY ACTIVITIES IN UKRAINE</b>	
Vita STROKAL, Maryna LADYKA, Yevgeniy BEREZHNIAK .....	1076
<b>SOIL PROFILES IN GATACKO POLJE IN EASTERN REPUBLIC OF SRPSKA, BOSNIA AND HERZEGOVINA</b>	
Vesna TUNGUZ, Turgay DINDAROGLU, Bojana PETROVIC.....	1083
<b>SOIL FERTILITY IN EASTERN REPUBLIC OF SRPSKA, BOSNIA AND HERZEGOVINA</b>	
Vesna TUNGUZ, Turgay DINDAROGLU, Bojana PETROVIC.....	1090
<b>ASSESSMENT OF POTENTIAL TOXIC ELEMENTS IN SOILS OF THREE URBAN AREAS AROUND VOLOS (GREECE) INDUSTRIAL ZONE</b>	
Georgios CHARVALAS, Elpiniki SKOUFOGIANNI, Aikaterini MOLLA, Savvas PAPADOPOULOS, Evangelia CHATZIKIROU, Aris KYPARISSIS, Olga CHRISTOPOULOU .....	1097
<b>CIRCULAR ECONOMY MODEL IN AQUACULTURE SECTOR IN LATVIA</b>	
Agnese EIZENBERGA .....	1102
<b>RESEARCH OF ARSENIC (As) CONTENT IN AGRICULTURAL LAND</b>	
Vladimir SABADOŠ, Danijela ŽUNIĆ .....	1108
<b>ANIMAL HUSBANDRY .....</b>	<b>1115</b>
<b>THE EFFECTS OF HEAT STRESS ON THE EXISTENCE OF STAPHYLOCOCCI IN MILK PRODUCED BY DAIRY COWS</b>	
Nassima BOUHROUM, Halima OUADEH, Leyla SLATNA, EL Hassan LANKRI ...	1116
<b>MICROBIOLOGICAL SAFETY OF FEED IN 2023</b>	
Bojan GOLIĆ, Dragan KNEŽEVIĆ, Dragan KASAGIĆ .....	1121
<b>THE IMPACT OF AGE ON REPRODUCTIVE TRAITS OF SIMMENTAL CATTLE</b>	
Jelena NIKITOVIC, Tatjana KRAJISNIK, Stevan RAJIC .....	1127
<b>HISTOLOGICAL ANALYSIS OF STRIATED MUSCULATURE WITH THE PURPOSE OF ASSESSING MEAT QUALITY</b>	
Nadžida MLAĆO, Amela KATICA.....	1133
<b>INVESTIGATION OF THE EFFECTS OF DOMESTIC WATER BUFFALO, HUNGARIAN RACKA SHEEP AND HUNGARIAN GREY CATTLE GRAZING ON DIFFERENT WOOD-PASTURES IN HUNGARY</b>	
Attila FŰRÉSZ, Szilárd SZENTES, Dénes SALÁTA, Zsombor WAGENHOFFER, Gabriella FINTHA, László SIPOS, Ferenc PAJOR, Károly PENKSZA .....	1141
<b>INBREEDING IN LATVIAN DARKHEADED SHEEP BREED BY PEDIGREE DATA</b>	
Liga PAURA, Daina JONKUS .....	1147
<b>COMPARATIVE EFFICACY OF DIFFERENT ACARICIDAL COMPOUNDS AGAINST TICKS INFESTING BOVINE IN AND AROUND RAWALAKOT AZAD KASHMIR, PAKISTAN</b>	

Anisa MUSHTAQ, Murtaz UL HASAN, Asim SHAMIM, Muhammad ALI ABDULLAH SHAH, Muhammad ARIF ZAFAR, Muhammad SHOAIB , Abdul ASIM FAROOQ, Saif UR REHMAN, Aayesha RIAZ, Muhammad KAMRAN ..... 1153

**THE FUNCTIONAL CHARACTERISTICS OF STARCH IN GROWING PIG NUTRITION**

Bojan STOJANOVIĆ, Vesna DAVIDOVIĆ, Aleksandra IVETIĆ, Blagoje STOJKOVIĆ, Saša OBRADOVIĆ, Ivana GRUJIČIĆ ..... 1160

**EVALUATING THE INCLUSION OF PUMPKIN SEED CAKE IN DAIRY COWS DIET ON MILK PRODUCTION**

Jordan MARKOVIĆ, Tanja VASIĆ, Sanja ŽIVKOVIĆ, Nedeljko RACIĆ, Đorđe LAZAREVIĆ, Marija STEPIĆ, Milomir BLAGOJEVIĆ ..... 1167

**THE CONTENT OF UREA IN THE MILK OF SIMMENTAL COWS IN DIFFERENT SEASONS AND HOUSING SYSTEMS**

Mladen TRIVUNOVIĆ, Ksenija ČOBANOVIĆ, Mirko IVKOVIĆ, Denis KUČEVIĆ, Tamara PAPOVIĆ ..... 1173

**CLAW LESIONS IN DAIRY COWS IN SERBIA**

Milan NINKOVIĆ, Nemanja ZDRAVKOVIĆ, Nemanja JEZDIMIROVIĆ, Jovan BOJKOVSKI, Svetla ARSIĆ ..... 1180

**ANIMAL PRODUCTS AS FUNCTIONAL FOOD AND ALIGNING THEIR USE WITH EU REGULATIONS**

Aleksandra MILOŠEVIĆ, Milena MILOJEVIĆ, Suzana KNEŽEVIĆ, Maja DOŠENOVIC MARINKOVIĆ, Goran STANIŠIĆ ..... 1184

**THE INFLUENCE OF POLLUTANTS ON THE CHEMICAL AND MICROBIOLOGICAL QUALITY OF THE RIVER IBAR**

Saša OBRADOVIĆ, Raško STEFANOVIĆ, Nenad ĐORĐEVIĆ, Bojan STOJANOVIĆ, Vladimir ŽIVKOVIĆ ..... 1190

**APPLICATION OF AROMA IN THE NUTRITION OF ANIMALS WITH FUR-CHINCHILLA**

Saša OBRADOVIĆ, Raško STEFANOVIĆ, Nenad ĐORЂEVIĆ, Bojan STOJANOVIĆ, Siniša BERJAN ..... 1196

**CHARACTERISTICS OF HERZEGOVINIAN CHEESE IN SACK**

Snežana JOVANOVIĆ, Tanja VUČIĆ ..... 1202

**THE EFFECT OF GENOTYPE AND PROTEASE ENZYME ON THE WEIGHT AND PERCENTAGE OF MEAT CLASSES IN CHICKENS**

Vladimir DOSKOVIĆ, Snežana BOGOSAVLJEVIĆ-BOŠKOVIĆ, Zdenka ŠKRBIĆ, Miloš LUKIĆ, Bojan STOJANOVIĆ, Simeon RAKONJAC, Veselin PETRIČEVIĆ .. 1208

**EFFECTS OF ROOSTER PRESENCE IN FREE-RANGE SYSTEMS ON EGG QUALITY**

Ali AYGUN, Dogan NARINC ..... 1213

**COMPARISON OF THREE DIFFERENT LAYER GENOTYPES RAISED IN A FREE-RANGE SYSTEM FOR EGG PRODUCTION PERFORMANCE**

Ali AYGUN, Dogan NARINC, Hasan ARISOY ..... 1218

<b>RESPONSE OF BROILERS TO A FERMENTED PROTEIN SUPPLEMENTARY FEED</b>	
Canan KOP-BOZBAY, Ahmet AKDAĞ.....	1225
<b>LAYING PERFORMANCE OF LAYING HENS FED A FERMENTED PROTEIN SUPPLEMENTARY FEED</b>	
Canan KOP-BOZBAY, Ahmet AKDAĞ.....	1231
<b>COMPUTATIONAL FLUID DYNAMICS (CFD) APPLICATIONS IN LIVESTOCK BUILDINGS</b>	
Yasemin KOCAMAN, Selda UZAL SEYFİ.....	1236
<b>THE EFFECT OF DIFFERENT RAISING CONDITIONS ON LIVE WEIGHT AFTER SHEARING AND GREASY WOOL YIELD IN FAT TAILED EWES</b>	
Turgut AYGÜN.....	1242
<b>OCCUPATIONAL ACCIDENTS, DISEASES AND PREVENTION IN CROP AND ANIMAL PRODUCTION</b>	
Turgut AYGÜN.....	1248
<b>CURRENT SITUATION OF HORSE BREEDING AND POSSIBILITIES IN BİNGÖL PROVINCE OF TÜRKİYE</b>	
Turgut AYGÜN.....	1255
<b>EFFECT OF BUCKWHEAT ON PERFORMANCE, EGG QUALITY AND HATCHING CHARACTERISTICS IN JAPANESE QUAIL</b>	
Vildan KOÇBEKER, Abdoulaziz Hamisso MAMAN, Ali AYGUN, Cansu BULUT, Ahmet GUNES .....	1261
<b>A STUDY CONCERNING THE EVALUATION OF THE BALANCE OF ENERGY DURING EARLY LACTATION</b>	
Vildan KOÇBEKER.....	1268
<b>THE EFFECT OF EGG WEIGHT ON CHICK QUALITY AND HATCHING RESULTS IN BROILER BREEDERS</b>	
Zeynep YARDIM, Nilgün UZUN, Elif AĞÖREN, A. Emre KARABAŞ .....	1274
<b>RURAL DEVELOPMENT AND AGRO-ECONOMY .....</b>	<b>1281</b>
<b>CHALLENGES OF SUSTAINABLE INNOVATION IN THE ALGERIAN FOOD INDUSTRY</b>	
Nouara BOULFOUL , Wassila HANAFI, Malika MAHDID, Salima SALHI.....	1282
<b>ENHANCING ORGANIZATIONAL MANAGEMENT AND DECISION-MAKING THROUGH DASHBOARD IMPLEMENTATION : A CASE STUDY OF THE ALGERIAN RED MEAT COMPANY, ALVIAR</b>	
Nouara BOULFOUL , Malika MAHDID, Amina BOUKHORS.....	1290
<b>RISK ANALYSIS OF INVESTING IN GREENHOUSE VEGETABLE PRODUCTION</b>	
Radomir BODIROGA, Grujica VICO, Milica STOJANOVIĆ, Ljubiša ŠEVKUŠIĆ, Stana TEŠIĆ, Jovana SPASOJEVIĆ.....	1300

<b>BIODIVERSITY FINANCE – CURRENT FINANCIAL FLOWS AND FUNDING GAPS</b>	
Zorica GOLIĆ, Saša LALIĆ .....	1308
<b>CLIMATIC REGIONALIZATION AND RAINFED AGRICULTURAL PRODUCTION IN THE STATE OF MARANHÃO, BRAZIL</b>	
Erika Costa SOUSA, Jose de Jesus Sousa LEMOS, Ronaldo Haroldo Nascimento de MENEZES .....	1318
<b>RURAL DEVELOPMENT VIA TRANSFORMING AGRO-INDUSTRIAL WASTE DERIVED FROM BIOMASS TO SUSTAINABLE BIOENERGY</b>	
Agapi VASILEIADOU .....	1327
<b>DEVELOPING TECHNOLOGY TRANSFER OFFICES TO SUPPORT INNOVATION: A CO-DESIGN PROCESS INITIATIVE IN BURKINA FASO AND NIGER</b>	
Annarita ANTONELLI, Francesco NOTARANGELO, Damiano PETRUZZELLA, Daniel DORI, Bilampa GNOUMOU/THIOMBIANO, Feridjou Emilie Georgette SANON/OUATTARA, Halima OUMAROU DIADIE, Daouda ANZA, Abdoulaye SOULEYMANE.....	1333
<b>AGRI-FOOD RESEARCH IN CHAD</b>	
Hamid EL BILALI, Tarek BEN HASSEN, Sinisa BERJAN .....	1340
<b>CIRCULAR AGRI-FOOD SYSTEMS: INTEGRATING NEXT-GEN TECHNOLOGIES FOR ZERO-WASTE AGRICULTURE</b>	
Luma Abu ORABI .....	1350
<b>ANALYSIS OF PRODUCTION AND EXPORT OF MONTENEGRIN WINE WITH PARTICIPATION OF SMALL PRODUCERS</b>	
Nataša PEROVIĆ, Bojana RISTANOVIĆ.....	1355
<b>PERCEPTUAL ANALYSIS OF MOUNTAIN ZONE GOVERNANCE: CASE OF THE TICHOUKT MASSIF - MIDDLE ATLAS, MOROCCO</b>	
Youssef BELHAJ , Mohamed OUCHERROU .....	1362
<b>DYNAMICS OF OASIS AGRICULTURE IN MOROCCO: BETWEEN INSTITUTIONAL GOVERNANCE AND CLIMATIC CHALLENGES</b>	
Laila CHIABRI, Abdellah BENTAHAR, Aziz MESSOUSSI.....	1368
<b>THE ECONOMIC AND ENVIRONMENTAL COST OF CULTIVATING OIL PALMS IN PAKISTAN</b>	
Rahatullah KHAN, Junaid Alam MEMON, Muhammad ARFAN.....	1375
<b>SETTLEMENT OF YOUNG FARMERS: IDENTIFICATION OF MAIN CONSTRAINTS FROM THE RESULTS OF A SURVEY</b>	
Pedro REIS, Maria de Fátima Lorena de OLIVEIRA, Ana NOVAIS .....	1381
<b>STRATEGIC OBJECTIVES OF THE DEVELOPMENT OF ORGANIC AGRICULTURE IN THE REPUBLIC OF SERBIA</b>	
Dragana ĐURIĆ, Jelena DAMNJANOVIĆ, Jelena MASLOVARA JOVANOVIĆ.....	1388
<b>THE POWER OF RURAL TOURISM: INSIGHTS FROM LOCAL COMMUNITY IMPACT STUDIES</b>	

Dunja DEMIROVIĆ BAJRAMI, Adriana RADOSAVAC, Desimir KNEŽEVIĆ .....	1397
<b>THE INFLUENCE OF YIELD AND PRICE FROM THE CURRENT YEAR ON SOWING AREA OF POTATO IN FOLLOWING YEAR – A MULTIPLE LINEAR REGRESSION</b>	
Ljiljana DRINIĆ, Nebojša NOVKOVIĆ, Dragana NOVAKOVIĆ .....	1404
<b>CHALLENGES OF SETTING UP AN ORGANIC VEGETABLES COOPERATIVE IN SERBIA</b>	
Ratko BOJOVIC, Richard SIMMONS, Steve QUARRIE .....	1410
<b>COMMON AGRICULTURAL POLICY INCENTIVES FOR HEDGEROW MANAGEMENT: A SLOVENIAN CASE STUDY</b>	
Andreja BOREC, Tina LEŠNIK .....	1417
<b>GROSS PROFIT ANALYSIS OF LEGUME-CEREALS ROTATION</b>	
Zuhal KARAKAYACI.....	1424
<b>COST BENEFIT ANALYSIS OF BULB ONION PRODUCTION USING MANUAL HARVESTING SYSTEM IN THE SOUTHEAST REGION, USA</b>	
Esendugue Greg FONSAH, Ajay Kiran NEMALI, Guy HANCOCK, Bhabesh DUTTA, Subas MALLA .....	1430
<b>CONSIDERATIONS CONCERNING THE EVOLUTION OF AGRITOURISM GUESTHOUSES IN ROMANIA</b>	
Irina Adriana CHIURCIU, Elena SOARE, Carina DOBRE .....	1437
<b>FORESTRY AND AGRO-FORESTRY .....</b>	<b>1443</b>
<b>STUDY OF THE CORK QUALITY FROM <i>QUERCUS SUBER</i> L. IN NORTH-WEST ALGERIA</b>	
Mustapha El Habib FRIHI, Belkheir DEHANE .....	1444
<b>INFLUENCE OF <i>JUNIPERUS SABINA</i>, ALGINITE AND BIOREGULATORS ON SEED GERMINATION KINETICS OF ROZNAVA ORIGIN BEECH</b>	
Dina ELISOVETCAIA, Raisa IVANOVA, Jan BRINDZA .....	1451
<b>CULTURE OF THE SILVER FIR (<i>ABIES ALBA</i> MILL.) IN ORAVIȚA VALLEY, ROMANIA</b>	
Cornelia BUZATU-GOANȚĂ .....	1457
<b>THE POTENTIAL OF LAND FOR THE IMPLEMENTATION OF FOREST CLIMATE PROJECTS IN RUSSIA</b>	
Svetlana MORKOVINA, Denis KUZNETSOV .....	1465
<b>CHARACTERIZATION OF ANTAGONISM OF FUNGI <i>EPICOCCUM NIGRUM</i> AND <i>DIAPORTHE ERES</i></b>	
Aleksandar VEMIĆ, Sanja JOVANOVIĆ, Aleksandar LUČIĆ, Zlatan RADULOVIĆ, Katarina MLADENOVIĆ, Ljubinko RAKONJAC, Vladan POPOVIĆ .....	1471
<b>TAXONOMY AND PHYTOGEOGRAPHY ANALYSIS OF MEDICINAL PLANTS WITHIN MANAGEMENT UNIT „GOČ-SELIŠTE“</b>	
Dušan JOKANOVIĆ, Vesna NIKOLIĆ JOKANOVIĆ, Kristina ŽIVANOVIĆ, Nikola ĐORĐEVIĆ, Bojan TUBIĆ .....	1477

<b>VARIABILITY OF VESSELS AND WOOD RAYS DENSITY BY CONTAINER SEEDLINGS OF DIFFERENT SPECIES FROM QUERCUS GENERA</b>	
Dušan JOKANOVIĆ, Vesna NIKOLIĆ JOKANOVIĆ, Kristina ŽIVANOVIĆ, Marko MARINKOVIĆ, Filip JOVANOVIĆ.....	1483
<b>THE CONDITION OF THE TREE CROWNS ON ICP LEVEL I PLOTS IN AP VOJVODINA, SERBIA</b>	
Milan DREKIĆ, Leopold POLJAKOVIĆ - PAJNIK, Marina MILOVIĆ, Zoran GALIĆ, Branislav KOVAČEVIĆ, Predrag PAP.....	1488
<b>INVESTIGATION OF ENVIRONMENTAL CONDITIONS FOR THE DEVELOPMENT OF FIR DECAYING FUNGI</b>	
Miroslava MARKOVIĆ, Renata GAGIĆ – SERDAR, Ljubinko RAKONJAC.....	1495
<b>EFFECT OF COMMERCIAL BIOFERTILIZER ON THE GROWTH PARAMETERS OF TWO-YEAR-OLD NORTHERN RED OAK (<i>QUERCUS RUBRA</i> L.) SEEDLINGS</b>	
Sanja JOVANOVIĆ, Aleksandar VEMIĆ, Aleksandar LUČIĆ, Ljubinko RAKONJAC, Vladan POPOVIĆ .....	1501
<b>PLANTING ENERGY CROPS FOR BIOMASS PRODUCTION ON DEPOSOL SOIL</b>	
Saša PEKEČ, Saša ORLOVIĆ, Branislav KOVAČEVIĆ, Marina MILOVIĆ, Leopold POLJAKOVIĆ PAJNIK, Lazar KESIĆ, Nikola PERENDIJA .....	1506
<b>ANALYSIS OF FOREST DAMAGE: 20-YEAR STUDY IN SERBIA</b>	
Zoran PODUŠKA, Snežana STAJIĆ, Branka PAVLOVIĆ, Jovana CVETKOVIĆ .....	1511
<b>SILVICULTURE IN THE FUNCTION OF SUSTAINABLE FOREST MANAGEMENT - CASE STUDY OF SE „SRBIJAŠUME“</b>	
Zvonimir BAKOVIĆ, Vladimir VASIĆ, Branko KANJEVAC, Snežana STAJIĆ .....	1518
<b>SILVOPASTURE SYSTEM COMMUNITY LIVESTOCK PERMIT HOLDERS PERCEPTIONS AND CHALLENGES IN THE LIMPOPO PROVINCE, SOUTH AFRICA</b>	
Phokele MAPONYA, Kgosi MONGWAKETSI, Takalani TAHULELA, Nico OLIVIER .....	1525
<b>SILVOPASTURE SYSTEM COMMUNITY LIVESTOCK PERMIT HOLDERS FOOD SECURITY STATUS IN THE LIMPOPO PROVINCE, SOUTH AFRICA</b>	
Phokele MAPONYA, Kgosi MONGWAKETSI, Takalani TAHULELA, Nico OLIVIER .....	1534
<b>DETERMINATION OF ARBUSCULAR MYCORRHIZAL FUNGI SPECIES AT DIFFERENT DEPTHS OF SOIL BENEATH THE PURE SESSILE OAK (<i>Quercus petraea</i> [Matt.] Liebl.) CANOPIES WITHIN A TEMPERATE FOREST STAND</b>	
Melih ÖZTÜRK, Şahin PALTA, Eren BAŞ, Kamil ÇAKIROĞLU.....	1540
<b>AUTHOR INDEX.....</b>	<b>1547</b>

## EVALUATING CORROSION AND INCRUSTATION RISKS IN SELECTED WELLS IN DANUBE ALLUVIUM

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### Abstract

The determination of corrosion and incrustation risks in drainage wells is crucial for the maintenance of water extraction infrastructure, ensuring the required efficiency in groundwater levels maintaining, and safeguarding the functionality and longevity of water extraction systems. Analytical methods have long been established for assessing water aggressiveness, providing insights into the groundwater environment susceptibility to corrosion or incrustation of underground structures. The assessment of groundwater aggressiveness is facilitated by the Langelier (LSI) and Ryznar indices (RSI), while Johnson's Classification aids in understanding incrustation phenomena. Based on conducted groundwater sampling campaign in 2021, for four drainage wells in Danube alluvium, the risks of chemical corrosion and incrustation were calculated. Revealed results indicated that groundwater was characterised with oxygen level of 2.5 mg/l – 3.75 mg/l and redox values in range of 362 mV to 583 mV, with very variable iron content. Based on the negative LSI values, examined groundwater from B-1, B-2, and B-4 is undersaturated and tends to dissolve calcium carbonate. The calculated indices for B-3 suggest mild incrustation tendencies, indicating a propensity for carbonate precipitation. Groundwater in zone B-4 stood out as highly corrosive. Saturation indices indicated groundwater tendencies towards either calcium carbonate dissolution or precipitation. The choice of construction materials significantly influences corrosion susceptibility, necessitating pre-emptive measures. Understanding groundwater behavior regarding scaling and corrosion is pivotal for effective prevention and maintenance planning. By implementing tailored strategies, such as pH adjustment and corrosion-resistant coatings, the detrimental effects of scaling and corrosion can be mitigated, ensuring the longevity of drainage well systems and minimizing operational costs. Regular monitoring of groundwater chemistry enables timely interventions, fostering sustainable water management practices.

**Keywords:** *groundwater, corrosion, incrustation, Langelier, Ryznar.*

### Introduction

The most common processes that contribute to the inevitable reduction in yield, known as "well aging," after long-term exploitation, are corrosion and incrustation-chemical scaling. The inclination of water towards either can significantly impact the transmission and distribution of water, influencing both economic considerations and public health concerns (Derakhshannia et al., 2020; Lodha et al., 2023; Mokhtari et al., 2010). Corrosion represents a set of chemicals, electrochemical, and biochemical processes that lead to damage and physical degradation of well construction. The basic precondition for the occurrence of corrosion of well construction metal parts is high mineralization, as well as the activity of microorganisms in water, especially sulfate-reducing bacteria. Favorable factors for the formation and development of corrosion include increased acidity of groundwater, increased content of dissolved oxygen (>2 mg/l), sulfides (>1 mg/l), and humic acids (>50 mg/l), presence of organic acids, low water hardness, and increased mineralization (>1000 mg/l). Incrustation or

chemical scaling is the process of depositing insoluble materials in the pre-filter zone of the well or in the well's filtration construction itself; it distinguishes mechanical, chemical, biochemical, and biological scaling. Chemical scaling is the process of forming precipitates through chemical reactions, usually in the form of scale on the filtration construction and in the pre-filter zone of the well. Favorable factors for the formation and development of incrustation include increased alkalinity of groundwater, increased content of bicarbonates ( $>300$  mg/l), iron ( $>2$  mg/l), and manganese ( $>1$  mg/l), and oxygen content below the aerobic boundary of the environment ( $<0.5$  mg/l). Pumping of groundwater leads to disturbance of the natural chemical equilibrium of groundwater as a result of hydrodynamic disturbances in the aquifer, contact of groundwater with atmospheric oxygen, loss of dissolved gases, and deposition of corresponding substances. Filtration and pre-filter zone scaling are conditioned by disturbances in the carbonate equilibrium of water and redox processes in the narrow zone of the well. In order to predict examined groundwater aggressiveness, the Langelier and Ryznar number were calculated, and Johnson's Classification were estimated.

### Materials and methods

The settlement of Vinci in Serbia is located on the right bank of the Danube River. Situated in the coastal area, Vinci occupies terrain elevations ranging from 70 to 75 meters above sea level (mnm), with the majority of the settlement positioned between 71 and 73 mnm. The lower-lying section of the settlement, below 72 mnm, lies adjacent to the Danube River, extending westward towards the "Vinci" water source (supplying the Golubac municipality), and partly towards the north and northwest, reaching the lower terrains of the Vinci-Požarevac drainage system.

In 1994, a drainage system consisting of a total of 4 operational wells (B-1 to B-4) with their own pumps and outlets into the Danube River was constructed to protect the lowest parts of the terrain in the developed area of the Vinci settlement (Figure 1). Following their construction, the wells were equipped with hydraulic and electrical equipment and put into operation. The primary role of this drainage system is to safeguard the lower coastal area of the Vinci settlement from the increased Danube levels. For these wells the Langelier and Ryznar indices were calculated as well as Johnson's Classification was determined.



Figure 1 The locations of the monitored wells

### The Langelier index and Ryznar number

The Ryznar Stability Index (RSI) is a parameter used to assess the scaling potential and corrosiveness of water (Ryznar, 1944). It helps determine whether water tends to form scale (such as calcium carbonate deposits) or if it is corrosive. The Ryznar index is calculated using water analysis data, including parameters like pH, conductivity, total dissolved solids (TDS), calcium concentration, bicarbonate concentration, and water temperature. Calculated pHs value, corresponds to the equilibrium state in the solution of carbonate compounds. The pHs value represents the hydrogen ion index corresponding to the equilibrium saturation of groundwater with carbonic acid compounds. The pH value reflects the real concentration of hydrogen ions in groundwater when measured directly on-site. Conversely, pH values obtained in laboratory settings typically register higher than those measured directly on-site. In light of analyzed literature data (Shankar, 2014), a practical formula has been proposed to calculate pHs values as follows (Eq.1):

$$\text{pH}_S = 9,92 - \frac{t \left[ ^0 C \right]}{40} - \log \left[ \text{Ca}^{++} (\text{mg/l}) \right] - \log \left[ \text{HCO}_3^- (\text{mg/ekv/l}) \right] + 0,2 \log \left[ \text{S.O.} (\text{mg/l}) \right]$$

(1)

t - temperature of groundwater ( $^{\circ}\text{C}$ ),

S.O. - dry residue (mg/l)

In terms of the stability of carbonate systems in water, the Ryznar criterion or number is introduced, expressed as:

$$\text{RSI} = 2\text{pHS} - \text{pH}$$

RSI has only positive values, and the higher the values, the more corrosive the water is (Table 1).

Table 1. Ryznar index and water tendency towards incrustation or corrosion

RSI value	Indication of water aggressiveness
4-5	Highly incrustive
5-6	Mildly incrustive
6-7	Mildly incrustive or corrosive
7-7.5	Corrosive
7.5-9	Highly corrosive
>9	Extremely corrosive

The Ryznar index complements the Langelier Saturation Index (LSI), which also assesses water scaling potential. While Ryznar gives an indication of water aggressiveness, the LSI considers both scale and corrosion potential (Eq. 2).

$$\text{LSI} = \text{pH} - \text{pHS}$$

(2)

If the measured pH value is less than the calculated pHs value ( $\text{LSI} < 0$ ), the concentration of dissolved  $\text{CO}_2$  in groundwater is above equilibrium, and groundwater can dissolve carbonate compounds. If, however, the pH value is greater than the calculated pHs value ( $\text{LSI} > 0$ ), the concentration of dissolved  $\text{CO}_2$  in groundwater is below equilibrium, and calcium carbonate can precipitate from groundwater.

Values of LSI can be interpreted as follows:

- Positive value: water tends to precipitate calcium carbonate.
- Zero: water is in equilibrium.
- Negative value: water is undersaturated, tends to dissolve calcium carbonate.

Johnson's Classification is relevant to incrustation processes. This classification helps in understanding the propensity of groundwater to form incrustations, which can be valuable for various applications, including water supply management and infrastructure maintenance. According to Johnson's Classification the process of incrustation occurs in groundwater with the following characteristics: the water is alkaline, i.e., pH is > 7.5, carbonate content in water exceeds 300 mg/L, indicating the potential for calcium carbonate precipitation. The iron content in water is greater than 2 mg/L. The process of iron precipitation from water is intensified by biological processes - the action of iron bacteria. The manganese content in water is over 1 mg/L.

## Results and Discussion

Considering the results of water quality testing, pH values ranged from a minimum of 7.16 in zone B-1 to a maximum of 7.62 in zone B-4, ranging from neutral to slightly alkaline values (Table 2). Carbonates were not detected, except in zone B-4, manganese content was >1 mg/L, and iron content varied widely, from 0.3 mg/l to even 5.42 mg/l (in B-2 and B-4). Along with high manganese values in groundwater in all four examined zones, microbial-mediated precipitation of iron and manganese and the formation of iron and manganese incrustations (iron and manganese oxide-hydroxides) are likely. Based on the negative values of LSI calculated according to the above relation, groundwater in the zones of hydro-technical facilities B-1, B-2, B-3, and B-4 is undersaturated and tends to dissolve calcium carbonate. The LSI value in zone B-3, according to the parameters calculated from measurements conducted in 2021, tends towards carbonate precipitation (Table 3). Based on the Ryznar number, groundwater in zones B-1 and B-2 fell within the range of corrosive waters, while zone B-3 exhibited characteristics of both weakly incrustive and corrosive waters. Groundwater in zone B-4 stood out as highly corrosive.

Table 2. Selected groundwater quality parameters

Sampling site		B-1	B-2	B-3	B-4
Parameter	Unit	7.16	7.23	7.62	7.53
pH вредност	µS/cm	499	488	583	362
Ec	mV	371.5	394.2	363	389.9
DO	mgO <sub>2</sub> /l	2.5	3.59	3.75	2.96
Iron (II)	mg/l	0.72	0.4	0.21	0.26
Iron	mg/l	2.77	5.42	0.72	0.3
NH <sub>4</sub>	mgN/l	1.36	0.72	1.02	0.14
NO <sub>2</sub>	mgN/l	0.006	0.054	0.038	0.028
NO <sub>3</sub>	mgN/l	0.15	0.81	0.62	2.83
Cl	mg/l	17.6	17.44	17.49	11.07
SO <sub>4</sub>	mg/l	3.33	15.5	18.35	25.47
H <sub>2</sub> S	mg/l	<0.04	<0.04	<0.04	<0.04
TOC	mg/l	2.48	1.98	3.8	3.89
OP	mgP/l	0.218	0.34	0.08	0.262
TDS	mg/l	322	302	360	243
Total hardness	mg CaCO <sub>3</sub> /l	222.4	233.4	275.4	151.3

Table 3. Saturation indices in groundwater

Well	Redox (mV)	LSI	Ryznar
B-1	371.5	-0.1	7.4
B-2	394.2	-0.08	7.4
B-3	363	0.42	6.8
B-4	389.9	-0.10	7.7

Several factors contribute to the corrosion of well materials. High mineralization levels in groundwater, along with the presence of dissolved gases and organic matter, can accelerate corrosion processes. Additionally, the pH level of the groundwater plays a crucial role, as acidic conditions can promote the dissolution of metallic components. Common materials used in well construction, such as steel, cast iron, and galvanized steel, are prone to corrosion if not adequately protected. Corrosion can lead to structural deterioration, leakage, and ultimately, the failure of the well system. Therefore, it is essential to select corrosion-resistant materials or apply protective coatings to extend the lifespan of wells and ensure the reliability of groundwater supply systems. Furthermore, proper design and construction practices, along with regular inspection and maintenance, are essential for mitigating corrosion risks in well infrastructure. By understanding the factors influencing corrosion and employing appropriate preventive measures, the integrity and functionality of well systems can be preserved, ensuring the sustainable management of groundwater resources.

## Conclusion

Understanding the groundwater potential for chemical scaling and corrosion is important from several aspects: managing prevention of scaling, corrosion control, and maintenance planning. Groundwater with high mineral content, particularly calcium carbonate, can lead to scaling in drainage well systems. This scaling can clog pipes, reduce flow rates, and interfere with the operation of pumps and other equipment. By knowing the groundwater potential, appropriate measures can be taken to mitigate scaling, such as adjusting pH levels or installing filtration systems. By assessing the potential for corrosion based on groundwater characteristics such as pH, dissolved oxygen content, and mineral content, protective measures like corrosion-resistant coatings or cathodic protection systems can be implemented to prolong the lifespan of drainage well infrastructure. Understanding the potential for chemical precipitation and corrosion allows for proactive maintenance planning. Regular monitoring of groundwater chemistry parameters can help identify changes over time, allowing for timely interventions to prevent or mitigate scaling and corrosion issues. This proactive approach can minimize downtime, reduce repair costs, and ensure the efficient operation of drainage well systems.

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# AUTHOR INDEX

A. Emre KARABAŞ .....	1273
Aanuoluwapo AYOBAMI OWOLABI .....	959
Aayesha RIAZ .....	1152
Abdelghani TAHIR .....	572
Abdelghani TAHIRI .....	566
Abdelhadi AJERRA .....	566
Abdelhak RHOUMA .....	703, 711
Abdellah BENTAHAR .....	1367
Abdelmalek MAHROUG .....	572
Abdelmalk MAHROUG .....	566
Abdoulaye SOULEYMANE .....	1332
Abdoulaziz Hamissou MAMAN .....	1260
Abdul ASIM FAROOQ .....	1152
Adam STAROWICZ .....	986
Adelina KARASTOEVA .....	851
Adnand RAMADHI .....	499
Adrian IOSIF .....	265, 271, 278, 592
Adriana RADOSAVAC .....	312, 1396
Agapi VASILEIADOU .....	1326
Agnese EIZENBERGA .....	1101
Agron BUNJAKU .....	499
Ahmad KATBEH-BADER .....	560
Ahmet AKDAĞ .....	1224, 1230
Ahmet GUNES .....	1260
Aikaterini GEORGALA .....	549
Aikaterini MOLLA .....	1096
Ajay Kiran NEMALI .....	1429
Albena PENCHEVA .....	124
Aleksa ĐUKIĆ .....	431
Aleksa LIPOVAC .....	327
Aleksandar ĐIKIĆ .....	391
Aleksandar LUČIĆ .....	1470, 1500
Aleksandar PAUNOVIĆ .....	312
Aleksandar VEMIĆ .....	1470, 1500
Aleksandar VUKOVIĆ .....	291, 391, 1013
Aleksandra ANGELESKA .....	938
Aleksandra BENSA .....	876, 883
Aleksandra BULAJIĆ .....	660
Aleksandra GOVEDARICA-LUČIĆ .....	59, 338, 345
Aleksandra IVETIĆ .....	402, 1159
Aleksandra LEPOSAVIĆ .....	660
Aleksandra MILOŠEVIĆ .....	1183
Aleksandra MILUTINOVIC .....	683
Aleksandra PERČIN .....	876
Aleksandra ŠUŠNjar .....	624, 649, 655
Aleksandrs ADAMOVICS .....	907
Aleksej LUKIN .....	108
Alexandr PCHELNICKOV .....	285
Alhassane ILLIAS .....	973
Ali AYGUN .....	1212, 1217, 1260
Ali SABIR .....	745
Alisa HADŽIABULIĆ .....	823
Alla BOROVSKAIA .....	253
Alma ČOLAK .....	823
Alma MEMIĆ .....	59
Alperen Kaan BÜTÜNER .....	719
Alvina AVAGYAN .....	53
Ama Souleymane ABOUBACAR .....	979
Amela KATICA .....	1132
Amina BOUKHORS .....	1289
Amira KHLIF .....	703, 711
Ana ANDĚLKOVÍC .....	619
Ana DOBREVA .....	858
Ana NIKOLIĆ .....	409
Ana NOVAIS .....	1380
Ana VUČEROVIĆ .....	613
Andra ZVIRBULE .....	913, 920
Andrej SELNEKOVIĆ .....	455
Andreja BOREC .....	1416
Aneta BUNTIĆ .....	772
Angelina PETKOVIĆ .....	89
Anisa MUSHTAQ .....	1152
Anita AUZINA .....	913, 920
Anna UHRINOVÁ .....	697
Annarita ANTONELLI .....	1332
Ansar KHOURI .....	239
Antonela MUSA .....	811, 830
Antonije ŽUNIĆ .....	624, 655
Antonije ŽUNIĆ .....	649
Arantzazu SANTAMARIA-ECHART .....	578
Arca AKBAŞ .....	731
Aris KYARISSIS .....	1096
Aristeidis K GEORGULIAS .....	485
Asim SHAMIM .....	1152
Athina TEGOU .....	193
Attila FŰRÉSZ .....	212
Attila FŰRÉSZ .....	1140
Aziz MESSOUSSI .....	1367
Bekri XHEMALI .....	499
Belkheir DEHANE .....	1443
Betim BRESILLA .....	499
Bhabesh DUTTA .....	1429
Bilampa GNOUMOU/THIOMBIANO .....	1332
Biljana BOŠKOVIĆ .....	298
Biljana PEĆANAC .....	513
Biljana SEVIC .....	303
Biljana ŠEVIĆ .....	327
Blagoje STOJKOVIĆ .....	1159
Bogdan NIKOLIĆ .....	628
Bojan DIMITRIJEVIĆ .....	478
Bojan GOLIĆ .....	513, 1120
Bojan GOSPODARIĆ .....	1058
Bojan KONATAR .....	1038
Bojan STOJANOVIĆ .....	1159, 1189, 1195, 1207
Bojan TUBIĆ .....	1476
Bojana LAPČEVIĆ .....	436
Bojana PETROVIC .....	1082, 1089
Bojana RISTANOVIĆ .....	1354
Boris BRKIĆ .....	366, 638
Boris PISINOV .....	628
Borislav PETKOVIĆ .....	64
Borislava BACHIYSKA .....	169
Boško GAJIĆ .....	352
Bouchra CHEBLI .....	566

Bozhidar KYOSEV .....	124
Branislav DRAŠKOVIĆ .....	817
Branislav KOVAČEVIĆ .....	1487, 1505
Branka GOVEDARICA .....	102, 108
Branka KRESOVIĆ .....	352
Branka PAVLOVIĆ .....	1510
Brankica KARTALOVIĆ .....	366, 378, 638
Brankica PEŠIĆ .....	613
Branko KANJEVAC .....	1517
Canan KOP-BOZBAY .....	1224, 1230
Cansu BULUT .....	1260
Carina DOBRE .....	1436
Carla CEDROLA .....	232
Ćerima ZAHIROVIĆ SINANOVIĆ .....	59
Christos ARAMPATZIS .....	894
Chrysi A. P APADIMITRIOU .....	889
Constantin LĂCĂTUŞU .....	278
Cornelia BUZATU-GOANȚĂ .....	1456
Cristina OLIVEIRA .....	578
Daina JONKUS .....	1146
Dalibor TOMIĆ .....	436
Damiano PETRUZZELLA .....	1332
Daniel DORI .....	1332
Daniela DJIKANOVIĆ .....	643
Daniil IAKOVLEV .....	285
Danijela JUNGIĆ .....	883
Danijela KONDIĆ .....	312
Danijela PETROVIĆ .....	811, 830
Danijela RISTIĆ .....	333, 409
Danijela ŠIKULJAK .....	619
Danijela ŽUNIĆ .....	1107
Danilo FURTULA .....	1038
Danka RADIĆ .....	643
Daouda ANZA .....	1332
Darko JEVREMOVIĆ .....	660
Darko MARKOVIĆ .....	817
Debasis MITRA .....	666
Dejan CVIKIC .....	303
Dejan CVIKIĆ .....	492
Dejan SKOČAJIĆ .....	999, 1006
Dejan ZEJAK .....	83
Dejana PANKOVIĆ .....	643
Dejana STANIĆ .....	519
Demetrios G. ROUPAKIAS .....	207
Dénes SALÁTA .....	212, 1140
Denis KUČEVIĆ .....	1172
Denis KUZNETSOV .....	1464
Denisa ŽUJO ZEKIĆ .....	823
Desimir KNEŽEVIĆ .....	312, 360, 1396
Diana KOPEVA .....	864
Dimitra-Despoina TOSILIANI .....	199
Dimitrios BARTZIALIS .....	193, 199
Dimitris M PAPAMICHAIL .....	485
Dina ELISOVETCAIA .....	1450
Divna SIMIĆ .....	402
Djamila YATTA-EL DJOUZI .....	42
Dogán NARINC .....	1212, 1217
Dominika GORNIK BUČAR .....	1058
Dorđe LAZAREVIĆ .....	1166
Dorđe MORAVČEVIĆ .....	327, 338
Dragan BOŽOVIĆ .....	360
Dragan GRČAK .....	312
Dragan KASAGIĆ .....	1120
Dragan KNEŽEVIĆ .....	513, 1120
Dragan RADOVANOVIC .....	1020
Dragan TOMOVIC .....	83
Dragan VUJIČIĆ .....	999, 1006
Dragana BOŠKOVIĆ .....	624, 649, 655
Dragana ĐURIĆ .....	1387
Dragana IGNJATOVIĆ-MICIĆ .....	333
Dragana LALEVIĆ .....	291, 391, 1013
Dragana MARINKOVIĆ .....	1020
Dragana MARISAVLJEVIĆ .....	619
Dragana NOVAKOVIĆ .....	1403
Dragana STAMENOV .....	416
Dragana STEVANOVIĆ .....	431
Dragana ŠUNJKA .....	624, 649, 655
Dragica MILOSAVLJEVIĆ .....	338, 345
Dragoslav ĐOKIĆ .....	291
Dunja ČOSIĆ .....	537
Dunja DEMIROVIĆ BAJRAMI .....	1396
Durđa PETROV .....	999, 1006
Dušan ČULUM .....	649
Dušan JOKANOVIĆ .....	1476, 1482
Dušan NIKOLIĆ .....	397, 402
Dušan UROŠEVIĆ .....	360
Dženita KOCA .....	320
Edmond MAICAN .....	265, 271, 278, 592
Eirini METAXA .....	900
Ekavi ISARI .....	894
EL Hassan LANKRI .....	1115
Elena LUTCAN .....	253
Elena PETROVIĆ .....	537, 543
Elena SOARE .....	1436
Elena TOFAN-DOROFEEV .....	947
Eleni WOGIATZI .....	193
Elif AĞÖREN .....	1273
Elizabeta DIMITRIESKA STOJKOVIKJ .....	938
Elma TEMIM .....	823
Elmira SALJNICKOV .....	397
Elpiniki SKOUFOGIANNI .....	1096
Elsa RAMALHOSA .....	578
Emina ADEMOVIĆ .....	823
Enika GREGORIĆ .....	320
Eren BAŞ .....	1539
Erika Costa SOUSA .....	1317
Esendugue Greg FONSAH .....	1429
Esin SERT .....	725
Evaggelia CHATZIKIROU .....	1096
Evgeniya VALCHINOVA .....	124
Fadi Sami KARAM .....	472
Fatma Zohra HAMDANI .....	505
Ferdinando BALDACCHINO .....	555
Ferenc PAJOR .....	212, 1140
Ferhan K. SABIR .....	738
Ferhan SABIR .....	745
Feridjou Emilie Georgette SANON/OUATTARA .....	1332
Filip JOVANOVIĆ .....	1482
Filipe LEMA .....	578

Filomena BARREIRO .....	578
Fjolla AVDYLAJ .....	555
Flutura LAMAJ .....	555
Francesco NOTARANGELO .....	1332
Francis DUCHIRON .....	807
Funmilayo MARY OLOYEDE .....	959
Gabriella FINTHA .....	212, 1140
Galina JEVĐENOVIC .....	772
Gazmend GJINOVCI .....	499
Georgi DIMITROV .....	530
Georgios CHARVALAS .....	1096
Gergana DESHEVA .....	124
Goran JAĆIMOVIĆ .....	383
Goran STANIŠIĆ .....	1183
Gordana ĐORĐEVIĆ .....	765
Gordana MATOVIĆ .....	320
Gorica CVIJANOVIĆ .....	478
Greta MALAJ .....	798
Grigor ZEHIROV .....	472
Grujica VICO .....	1299
Gulnari CHKHUTIASHVILI .....	189
Gunta GRINBERGA-ZALITE .....	913, 920
Guy HANCOCK .....	1429
Halima OUADEH .....	1115
Halima OUMAROU DIADIE .....	1332
Hamid EL BILALI .....	1339
Hasan ARISOY .....	1217
Helena BREKALO .....	811, 830
Hüseyin SENSOY .....	1064
Ibrahim AL-JBOORY .....	560
Igor ĐURĐIĆ .....	89, 102, 108
Igor ILJOVSKI .....	246
Igor VUKELIĆ .....	643
Ile CANEV .....	246
Ilija KOMLJENOVIC .....	64
Ilinka PEĆINAR .....	643
İlyas BOLAT .....	1064
Imane KRAOUCHE .....	505
Ioanna CHOMATA .....	889
Ioannis KALAVROUZIOTIS .....	894
Ioannis N. XYNIAS .....	207
Ippolitos GINTSIUDIS .....	199
Irina Adriana CHIURCIU .....	1436
Irina MARINA .....	298
Isabel CALHA .....	584
Isidora SIMOVIĆ .....	999, 1006
İsmail Alper SUSURLUK .....	719
Ivan TUPAJIĆ .....	327
Ivana GAJIĆ .....	690, 693
Ivana GRUJIĆ .....	1159
Ivana KOLLÁROVÁ .....	448
Ivana MEZEYOVÁ .....	448, 455, 465
Ivana TIRDIL'OVÁ .....	465
Ivana TOŠIĆ .....	836
Ivana ZIVKOVIC .....	303
Jamal Ali Mohamed EHIDADAN .....	927, 932
Jan BRINDZA .....	1450
Ján MEZEY .....	448, 455, 465
Janij KAVČIČ .....	1058
Janis MILLERS .....	907
Jasenka ČOSIĆ .....	537, 543
Jasmina BAĆIĆ .....	360
Jelena DAMNjanović .....	492, 1387
Jelena DRAGIŠIĆ MAKsimović .....	338, 345
Jelena EĆIMOVIĆ .....	624, 649, 655
Jelena IVANović .....	402
Jelena MAKsimović .....	772
Jelena MASLOVARA JOVANOVIĆ .....	1387
Jelena MRMOŠANIN .....	690, 693
Jelena NIKITOVIC .....	1126
Jelena STOJILJKOVIC .....	303
Jelena STOJILJKOVIĆ .....	327
Jelena VANČETOVIĆ .....	333
Jelena VISKOVIĆ .....	383
Jelena VUKADINović .....	409
Jonel SUBIĆ .....	298
Jordan MARKOVIĆ .....	660, 666, 1166
Jose de Jesus Sousa LEMOS .....	1317
Jovan BOJKOVSKI .....	1179
Jovan PAVLOV .....	180
Jovana CVETKOVIĆ .....	1510
Jovana HRUSTIĆ .....	613
Jovana SPASOJEVIĆ .....	1299
Júlia FABIANOVÁ .....	465
Junaid Alam MEMON .....	1374
Jurica ŠKRLEC .....	883
Kamenko BRATKOVIĆ .....	360
Kamil ÇAKIROĞLU .....	1539
Karolina VRANDEČIĆ .....	537, 543
Károly PENKSZA .....	212
Károly PENKSZA .....	1140
Katarina GAJIĆ .....	352
Katarina MATAN .....	876
Katarina MLADENović .....	1470
Katya UZUNDZHALIEVA .....	130, 174
Kgosi MONGWAKETSI .....	1524, 1533
Kristina ŽIVANOVIC .....	1476, 1482
Ksenija ČOBANOVIĆ .....	1172
Ksenija RADOTIĆ .....	643
Kyriakos D. GIANNOULIS .....	193, 199
Laila CHIABRI .....	1367
László SIPOS .....	212, 1140
Laura TADEVOSYAN .....	53
Layla NAIM .....	472
Lazar KESIĆ .....	1505
Lazar PEJIĆ .....	677
Leopold POLJAKOVIĆ - PAJNIK .....	1487
Leopold POLJAKOVIĆ PAJNIK .....	1505
Levan UJMAJURIDZE .....	189
Leyla SLATNA .....	1115
Lidija MILENKOVIĆ .....	1013
Liga PAURA .....	1146
Lilyana KOLEVA .....	530
Ljiljana BRBAKLIĆ .....	383
Ljiljana DRINIĆ .....	1403
Ljiljana TANASIĆ .....	817
Ljubinko RAKONJAC .....	1038, 1470, 1494, 1500
Ljubiša ŠEVKUŠIĆ .....	1299
Ljupco ANGELOVSKI .....	938
Lobna HAJJI-HEDFI .....	703, 711

Lubna NASIR EDDEEN .....	239
Lucia UNGVARSKÁ MALUČKÁ .....	697
Luma Abu ORABI .....	1349
Lutvija KARIĆ .....	59
Lydia BOUAMRA .....	42
Magdalena KNEŽEVIĆ .....	772
Mahamadou MOUSSA DIT KALAMOU .....	979
Maharazu A. YUSUF .....	979
Maja DOŠENOVIĆ MARINKOVIĆ .....	1183
Maja SUDIMAC .....	327
Maja ŽIVANOVIĆ .....	613
Malika MAHDID .....	1281, 1289
Maman Sani ABDOU BABAYE .....	973
Manal BELATIK .....	953
Manol DESHEV .....	124
Marcin ZIELIŃSKI .....	986
Margarita HARUTYUNYAN .....	53
Mária BREZÁNIOVÁ .....	448
Maria de Fátima Lorena de OLIVEIRA .....	1380
Maria DE FÁTIMA OLIVEIRA .....	584
Maria do Céu FIDALGO .....	578
Maria Halana FLORINDO DA SILVA .....	578
Mariana IONESCU .....	599
Marija BAJAGIĆ .....	478
Marija ČOSIĆ .....	327
Marija KOSTADINOVIC .....	333
Marija MILOSAVLJEVIĆ .....	1046
Marija PEROVIĆ .....	1026, 1032, 1053
Marija SRBINOSKA .....	677
Marija STEPIĆ .....	1166
Marijana DUGALIĆ .....	352
Marina DERVIŠEVIĆ .....	772
Marina HOVHANNISYAN .....	53
Marina JOVKOVIĆ .....	772
Marina MILOVIĆ .....	1487, 1505
Marko KOSTIĆ .....	372
Marko MARINKOVIĆ .....	1482
Marko PETKOVIĆ .....	76
Maryna LADYKA .....	1075
Maša BUĐEN .....	366, 372, 378, 638
Mate BOBAN .....	811, 830
Mauro PAGANO .....	232
Mehdi MAQBOOL .....	259
Mehmet ARSLAN .....	786, 792
Melih ÖZTÜRK .....	1539
Melpomeni NIKOU .....	485, 900
Mihajlo LIZDEK .....	83
Mihajlo MARKOVIĆ .....	64
Milan BIBERDŽIĆ .....	291, 391, 1013
Milan BRANKOV .....	352
Milan DREKIĆ .....	1487
Milan GLIŠIĆ .....	817
Milan MIROSAVLJEVIĆ .....	383
Milan NINKOVIĆ .....	1179
Milan UGRINOVIĆ .....	492
Milanka MAŠANOVIC .....	519
Milena MILJKOVIĆ .....	690, 693
Milena MILOJEVIĆ .....	1183
Milena MITEVA .....	858
Milena SIMIC .....	303
Milica BALOŠ .....	423
Milica LJUBOJEVIC .....	70
Milica LUČEV .....	333
Milica MIHAJLOVIĆ .....	613
Milica STOJANOVIC .....	59, 338, 345, 1299
Milomir BLAGOJEVIĆ .....	1166
Milomirka MADIĆ .....	436
Milorad VESELINOVIĆ .....	1046
Miloš DUGALIĆ .....	628
Miloš LUKIĆ .....	1207
Miloš MARJANOVIĆ .....	436
Miloš PAJIĆ .....	298
Miloš PETROVIĆ .....	624
Milosav GRČAK .....	312
Mima ILCHOVSKA .....	138
Miodrag TOLIMIR .....	352
Mira MILINKOVIĆ .....	772
Mirela MATKOVIĆ STOJŠIN .....	312, 360
Mirjam VUJADINOVIC MANDIĆ .....	320
Mirjana JOVOVIĆ .....	76, 845
Mirjana OCOKOLJIĆ .....	999, 1006
Mirjana RADOVIC .....	83
Mirjana VUKOSAVLJEV .....	366, 372, 378, 638
Mirko IVKOVIĆ .....	1172
Mirko KULINA .....	83
Miroslav LALOVIĆ .....	817
Miroslav ŠLOSÁR .....	448, 465
Miroslava MARKOVIĆ .....	1038, 1046, 1494
Mladen PETROVIĆ .....	478, 765
Mladen TRIVUNOVIĆ .....	1172
Mohamed ALOUANI .....	566, 572
Mohamed KHARSSI .....	42
Mohamed OUCHERROU .....	1361
Mohsen JANMOHAMMADI .....	218, 222, 226
Momir ALVIROVIĆ .....	765
Monika SABEVA .....	864
Muhammad ALI ABDULLAH SHAH .....	1152
Muhammad ARFAN .....	1374
Muhammad ARIF ZAFAR .....	1152
Muhammad KAMRAN .....	1152
Muhammad SHOAIB .....	1152
Murtaz UL HASAN .....	1152
Mustapha El Habib FRIHI .....	1443
Mustapha OUATMANE .....	953
Nabila BELHAMICHE .....	807
Nada JELIĆ .....	683
Nadžida MLAĆO .....	1132
Naima AIT AABD .....	566, 572
Naima CHABB .....	572
Namira MHAMEDI BOUZINA .....	505
Naser SABAGHNIA .....	218, 222, 226
Nassima BOUHROUM .....	1115
Natalija KRAVIĆ .....	409
Nataliya PETROVSKA .....	138, 145
Natalya PETROVSKA .....	180
Nataša LJUBIĆIĆ .....	366, 372, 378, 638
Natasa MARIC .....	70, 118
Natasa PEROVIĆ .....	1354
Nebojša NOVKOVIĆ .....	1403
Nedeljka ELEZ .....	89

Nedeljko RACIĆ .....	1166
Nedyalka YORDANOVA .....	151
Neluş-Evelin GHEORGHITĂ .....	278, 599, 605
Nemanja JEZDIMIROVIĆ <sup>i</sup> .....	1179
Nemanja ZDRAVKOVIĆ .....	1179
Nenad DELIC .....	180
Nenad ĐOREĐEVIĆ .....	1189, 1195
Nenad MALIĆ .....	836
Nenad PAVLOVIĆ .....	436
Nenad ŽIVKOVIĆ .....	431
Nevena ČULE .....	1046
Nevena STEVANović .....	366, 372, 378, 638
Nevenka GALEČIĆ .....	999, 1006
Nicholaos G. DANALATOS .....	193, 199
Nico OLIVIER .....	1524, 1533
Nicolae-Valentin VLĂDUȚ .....	265, 278
Nicolai VANICOVICI .....	253
Nicoleta UNGUREANU .....	278
Nidal SHABAN .....	472
Nikola ĐORĐEVIĆ .....	423, 1476
Nikola GRČIĆ .....	409
Nikola LAĆARAC .....	624
Nikola LJILJANIĆ .....	397
Nikola PERENDIJA .....	1505
Nikola STANKOVIĆ .....	366, 372, 378, 638
Nikolay MINEV .....	151, 157
Nikolina JURKOVIĆ BALOG .....	876
Nilgün UZUN .....	1273
Nimet Sema GENÇER .....	725
Noosheen ZAHID .....	259
Nouara BOULFOUL .....	49, 1281, 1289
Nuray BASER .....	649
Odeta POCIENE .....	779
Ola SHALBAK .....	239
Olga CHRISTOPOULOU .....	1096
Olga IONITA .....	947
Olivera ŠUŠA .....	1013
Omaima BARGOUGUI .....	703, 711
Ornela SHOSHI .....	798
Özlem KIZILIRMAK ESMER .....	731
Panagiotis TZIACHRIS .....	900
Paula BAPTISTA .....	578
Paulina RUSANOWSKA .....	986
Pedro REIS .....	584, 1380
Penka VULCHINKOVA .....	169
Petra ŠEVČÍKOVÁ .....	455
Petros KOKKINOS .....	894
Phokele MAPONYA .....	1524, 1533
Predrag BRKOVIĆ .....	431
Predrag PAJIĆ .....	1053
Predrag PAP .....	1487
Rachid BOUHARROUD .....	566, 572
Rade JOVANOVIĆ .....	402
Rade STANISAVLJEVIĆ .....	291
Radia. AYADI .....	42
Radmila CRCEVA NIKOLOVSKA .....	938
Radomir BODIROGA .....	345, 1299
Radomir LJUPKOVIĆ .....	690
Rahatullah KHAN .....	1374
Raheel ANWAR .....	259
Raisa IVANOVA .....	253, 1450
Raluca-Adriana ZOTA .....	605
Rasa SLINKSIENE .....	779
Raško STEFANOVIĆ .....	1189, 1195
Ratko BOJOVIC .....	1409
Raya BALAYAN .....	53
Redouan QESSAOUI .....	566, 572
Renata GAGIĆ – SERDAR .....	1494
Renata GAGIĆ-SERDAR .....	1038
Richard SIMMONS .....	1409
Risto UZUNOV .....	938
Roberto TOMASONE .....	232
Ronaldo Haroldo Nascimento de MENEZES <sup>3</sup> .....	1317
Rosa. SLAMANI .....	42
Şahin PALTA .....	1539
Said BENALLAOUA .....	807
Said. BOUDEFFEUR .....	42
Saif UR REHMAN .....	1152
Salahddine CHAFIKI .....	566, 572
Salima SALHI .....	1281
Samar DALI .....	703, 711
Sana. EZZOUAOUI .....	42
Sancar BULUT .....	786, 792
Sanda MAICAN .....	271, 592
Sandra MEDić .....	811, 830
Sandra MOJSOVA .....	938
Sanja ĐUROVIĆ .....	628
Sanja JOVANOVIĆ .....	1470, 1500
Sanja LAZIĆ .....	655
Sanja LAZIĆ .....	649
Sanja MIKIĆ .....	383
Sanja ŽIVKOVIĆ .....	660, 666, 1166
Saša BARAĆ .....	291, 391, 1013
Saša LALIĆ .....	108, 1307
Saša OBRADOVIĆ .....	1159, 1189, 1195
Saša ORLOVIĆ .....	1505
Saša PEKEĆ .....	1505
Sasho SABEV .....	851
Savvas PAPADOPoulos .....	193, 1096
Selda UZAL SEYFİ .....	1235
Selena ĆEVRIZ .....	845
Sevil UNAL .....	738, 745
Shpend SHAHINI .....	499
Simeon RAKONJAC .....	1207
Simonida ĐURIĆ .....	416
Sinisa BERJAN .....	1339
Siniša BERJAN .....	1195
Slađan STANKOVIĆ .....	397, 402
Slađana JANKOVIĆ .....	89
Sladjana PETRONIC .....	70, 118
Slavica ČABRILO .....	683
Slavica VUKOVIĆ .....	649
Slavica VUKOVIĆ .....	624, 655
Slavisa MILOVANOVIC .....	83
Slavko GRGIĆ .....	537
Slobodan KRNJAJIĆ .....	397
Snežana BOGOSAVLJEVIĆ-BOŠKOVIĆ .....	1207
Snežana BRKLJAČ .....	95
Snežana ĐORĐEVIĆ .....	423, 442
Snežana JANKOVIĆ .....	402

Snežana JOVANOVIĆ .....	1201
Snežana MLADENOVIĆ DRINIĆ .....	409
Snežana OLJAČA .....	442
Snežana STAJIĆ .....	1046, 1510, 1517
Sonja FILIPOVIĆ .....	666
Sorin-Stefan BİRİŞ .....	265
Sorin-Ştefan BİRİŞ .....	271, 592
Sorin-Ştefan BİRİŞ .....	278, 599, 605
Souheila OUANAS .....	807
Souleymane ISSA MALAN S .....	973
Srdan ŠEREMEŠIĆ .....	442
Stana TEŠIĆ .....	1299
Stanislav STAMATOV .....	174
Stefan KOVAČEVIĆ .....	619, 660
Stefan VULCHINKOV .....	180
Stefana UTVIĆ .....	402
Steva LEVIĆ .....	643
Stevan RAJIC .....	1126
Steve QUARRIE .....	31, 1409
Stjepan HUSNJAK .....	883
Subas MALLA .....	1429
Sultan ARSLAN-TONTUL .....	1070
Suzana KNEŽEVIĆ .....	1183
Suzana MITROVIĆ .....	1038, 1046
Suzana PAVLOVIĆ .....	492
Svetla ARSIĆ .....	1179
Svetlana KOSTADINOVA .....	151, 157
Svetlana MORKOVINA .....	1464
Szilárd SZENTES .....	212, 1140
Takalani TAHULELA .....	1524, 1533
Tamara PAPOVIĆ .....	1172
Tamara SIBER .....	537, 543
Tanja JAKIŠIĆ .....	102
Tanja MIJATOVIĆ .....	102, 108
Tanja PEROVIĆ .....	59
Tanja VASIĆ .....	660, 666, 1166
Tanja VUČIĆ .....	1201
Tarek BEN HASSEN .....	1339
Tatiana KRIUKOVA .....	108
Tatiana SCHREINER .....	578
Tatjana IVANOVIVIĆ .....	312
Tatjana KRAJISNIK .....	1126
Thaer YASEEN .....	560
Theano B. LAZARIDOU .....	207
Theodoros MAVROMATIS .....	485
Tijana BANJANIN .....	95
Tijana DUDIĆ .....	628
Tijana UROŠEVIĆ .....	672, 677
Timea HAJNAL JAFARI .....	416
Tina LEŠNIK .....	1416
Toni GALIĆ .....	830
Tsotne SAMADASHVILI .....	189
Tufan GÜRAY .....	752, 758
Turgay DINDAROGLU .....	1082, 1089
Turgut AYGÜN .....	1241, 1247, 1254
Ulku Dilek UYSAL .....	758
Ülkü Dilek UYSAL ÖZDEMİR .....	752
Una MATKO .....	836
Valentina VALKOVA .....	138, 163, 169
Valya BACHIYSKA .....	138
Vanya SLAVOVA .....	130
Vassilis G ASCHONITIS .....	485
Veaceslav GHENDOV .....	947
Vedran TOMIĆ .....	397
Vera KOLEVA .....	870
Vera POPOVIĆ .....	372
Verica MILOVIĆ .....	76
Vesela RADOVIĆ .....	397
Veselin PETRIČEVIĆ .....	1207
Veselin SEVOV .....	157
Veselina MASHEVA .....	130, 174
Veselinka ZEČEVIĆ .....	312, 360
Vesna DAVIDOVIĆ .....	1159
Vesna DRAGICEVIC .....	303
Vesna DRAGIČEVIĆ .....	442
Vesna KRNJAVA .....	423
Vesna MILIĆ .....	102, 108
Vesna NIKOLIĆ JOKANOVIĆ .....	1476, 1482
Vesna OBRADOVIĆ .....	1026, 1032, 1053
Vesna POČUĆA .....	320
Vesna RADOJIČIĆ .....	672, 677
Vesna TUNGUZ .....	1082, 1089
Viktoria TRIFONOVA .....	858
Vildan KOÇBEKER .....	1260, 1267
Violeta ANĐELKOVIĆ .....	409
Violeta MANDIĆ .....	423
Violeta MICKOVSKI STEFANOVIĆ .....	431
Violina ANGELOVA .....	870
Višnja SIKIMIĆ .....	683
Vita STROKAL .....	1075
Vladan JOVANOVIĆ .....	628
Vladan POPOVIĆ .....	1470, 1500
Vladeta STEVOVIĆ .....	436
Vladimir AĆIN .....	383
Vladimir ČIRIĆ .....	1020
Vladimir DOSKOVIĆ .....	1207
Vladimir FILIPOVIĆ .....	431
Vladimir SABADOŠ .....	1107
Vladimir VASIĆ .....	1517
Vladimir ŽIVKOVIĆ .....	1189
Vladimíra HORČINOVÁ SEDLÁČKOVÁ .....	53
Vladimira ŽUNIĆ .....	416
Vojin CVIJANOVIĆ .....	478, 765
Vojkan MILJKOVIĆ .....	690, 693
Vojo RADIĆ .....	64
Vuk MAKSIMOVIĆ .....	338, 345
Wassila HANAFI .....	49, 1281
Yana KOLEVA .....	858
Yasemin KOCAMAN .....	1235
Yevgeniy BEREZHNIAK .....	1075
Youssef ABOU OBEID .....	472
Youssef BELHAJ .....	1361
Yuliana IORDANOVA .....	145
Zahidenur TURAN .....	745
Zdenka GIREK .....	492
Zdenka ŠKRBIĆ .....	1207
Zeljana DOKNIC .....	118
Zeljko DOLIJANOVIC .....	303
Željko DOLIJANOVIC .....	442
Zeynep YARDIM .....	1273

Zhelyazko VULCHINKOV .....	169, 180	Zoran STOJANOVIĆ .....	765
Zlatan RADULOVIĆ .....	1046, 1470	Zoranka MALEŠEVIĆ .....	76, 845
Zlatko ARSOV .....	246	Zorica GOLIĆ .....	1307
Zoran DIMOV .....	246	Zorica RANKOVIĆ VASIĆ .....	95
Zoran DINIĆ .....	338	Zorica SREDOJEVIĆ .....	352
Zoran GALIĆ .....	1487	Zsombor WAGENHOFFER .....	212, 1140
Zoran MALIČEVIĆ .....	64	Zuhal KARAKAYACI .....	1423
Zoran PODUŠKA .....	1510	Zvonimir BAKOVIĆ .....	1517
Zoran S. ILIĆ .....	1013		