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# Evaluation of Seasonal Changes in the Content of Trace Elements in *Satureja kitaibelii*

Dragoljub Miladinović<sup>a</sup> (b), Marija Dimitrijević<sup>a</sup> (b), Jelena Mrmošanin<sup>b</sup>, Marija Marković<sup>c</sup>, and Aleksandra Pavlović<sup>b</sup>

<sup>a</sup>Department of Pharmacy, Faculty of Medicine, University of Niš, Niš, Serbia; <sup>b</sup>Department of Chemistry, Faculty of Science and Mathematics, University of Niš, Niš, Serbia; <sup>c</sup>Department of Biology and Ecology, Faculty of Science and Mathematics, University of Niš, Niš, Serbia

#### ABSTRACT

This study aimed at combined chemical and chemometric analysis of seasonal variation in trace element (TE) contents (B, Si, Cr, Mn, Ni, Cu, and Zn) in Satureja kitaibelii Wierzb. ex Heuff. (savory) and its growing soil, with emphasis on potential aspects of health promotion. The lowest contents of the studied elements in the soil, except for silicon, were recorded in the vegetative stage. In the plant, the boron content is the highest:  $10.5-14.9 \text{ mg kg}^{-1}$  and the chromium content is the lowest:  $0.17-1.2 \text{ mg kg}^{-1}$ . The highest values of soilto-plant transfer factor (TF) were recorded for five elements (except Si and Cr) in the vegetative stage. Savory can be considered an accumulator of boron. The statistical and chemometric analysis confirmed a variation in TE content of studied soil and plant samples. This study revealed that savory from Serbia to be a potential source of valuable TEs. A significant percentage of daily intake of B, Cr, and Ni, can be provided with three cups of tea per day of plants collected in the vegetative and flowering stages.

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#### **KEYWORDS**

Chemometric analysis; dietary intake; *Satureja kitaibelii*; seasonal changes; trace element

# Introduction

Trace elements (TEs) take part in vital biochemical and physiological functions, which are necessary for life maintenance. Some TE are essential for plants and animals, and they are responsible for the medicinal properties of herbs (Jungová et al. 2022). According to the World Health Organization, traditional medicine based on herbal remedies represents an important part of health services (WHO 2013). Related to these facts, medicinal plants are also used as food ingredients to combine adequate amounts of nutrients and chemical elements that are essential to a normal diet.

*Satureja kitaibelii* Wierzb. ex Heuff. is one of the most popular herb in Serbia (also known as Rtanj tea), used as a culinary plant, as well as tea in traditional medicine (Miladinović et al. 2014, 2022). While the chemical composition of secondary metabolites and their biological activities have been investigated (López-Cobo et al. 2015; Stanojković et al. 2013; Dodoš et al. 2019; Gopčević et al. 2019; Dorđević et al. 2014), there is not enough data on TE content in *S. kitaibelii* along with soil-to-plant transfer

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factors (TFs). Having in mind these facts, also given importance of savory as a favorite medicinal and spice plant, the objectives of this study were chemical and chemometric analysis of seasonal variation in seven TE contents (B, Si, Cr, manganese [Mn], Ni, copper [Cu], and Zn), in the soil and *S. kitaibelii*, at the Kravlje village, southeastern Serbia, with emphasis on potential aspect of health promotion. To the best of our knowledge, this article is the first one about the content of TEs in *S. kitaibelii*. The concentrations of elements were determined by inductively coupled plasma-optical emission spectrometry (ICP-OES), in combination with chemometric analysis.

# **Material and methods**

# **Chemicals and reference materials**

All reagents were analytical-reagent grade. Nitric acid (65%), hydrochloric acid (36%), and hydrogen peroxide (30%) were purchased from Merck (Darmstadt, Germany). For all dilutions, deionized water was used.iTEVA software from Thermo Scientific (Cambridge, UK) was used to collect and analyze the data (Pavlović et al. 2020). Multielement standard solution IV of the microelements Al, As, Ba, Be, B, Cd, Cr, Co, Cu, Fe, Pb, Mn, Ni, Se, Tl, V, and Zn, standard solution III of the macroelements Ca, K, Mg, and Na, as well as individual standard solutions of Si, P, and Hg (Trace CERT, Fluka Analytical, Switzerland) were used for calibration.

# Sample collection

The aerial parts of S. kitaibelii Wierzb. ex Heuff. family Lamiaceae were collected from a natural population at the Kravlje village. The village Kravlje is in the Miljkovačka gorge near city of Niš in the hilly-mountainous region of Southeastern Serbia. The importance of the locality Kravlje, from which the samples of S. kitaibelii were collected, is reflected in the favorable micro climatic conditions for plant growth. The populations of plant species S. kitaibelii can be found on the south-facing rocky mountainsides, on the rocky grounds as habitat, in which the temperature and insolation are more favorable for their survival (Miljković, Ranđelović, and Ranđelović 2012). Plant samples were collected during 2020 at three different stages of development: vegetative stage (June, M1); flowering stage (July, M2 and August, M3); after flowering stage (September, M4; October, M5 and November, M6). The plants were collected on the fifteenth in the months mentioned. Dr. Marija Marković did identification of plant material, and the voucher specimen (accession number 13220) is deposited at the Herbarium of the Department of Biology and Ecology, Faculty of Science and Mathematics, University of Niš (Herbarium Moesiacum Niš - HMN). Basic characteristics of the locality are given in Table 1.

Table 1. Basic characteristics of the locality Kravlje village.

H <sup>a</sup> (m)	N <sup>b</sup>	Ec	Substrate	Climate
332	$43^\circ~27'~16''$	$21^\circ~54^\prime~20^{\prime\prime}$	Quartz	Temperate continental
3 b.				

<sup>a</sup>Altitude; <sup>b</sup>Latitude; <sup>c</sup>Longitude.

Seven sample locations were selected, and seven TE contents were studied: B, Si, Cr, Mn, Ni, Cu, and Zn. The topsoil (0-20 cm) of the sample mixture, consisting of three small samples, was collected 10 m apart at each sampling point. Seven samples of *S. kitaibelii* in the same growth phase were taken at each sample site within  $20 \times 20 \times 20$  cm soil blocks, cut using a stainless steel spade. The soil dust and other materials in the savory samples were removed with a plastic brush, washed repeatedly with distilled water, and then stored in pre-cleaned polythene bags. The collected samples were then brought to the laboratory for further processing.

# Soil sampling and preparation

Each soil sample was carefully mixed and external materials, such as stones and pebbles were extracted. The sample was then heated in an electric oven at 60 °C until a constant weight was obtained. Dried soil samples are ground into fine powder. Weighted soil sample mass (1.00 g) was placed into an Erlenmeyer and treated with 16 mL mixture of conc. HCl and conc. HNO<sub>3</sub> (3:1) (v/v). The mixture was heated to 190 °C for about an hour, then 5 mL of H<sub>2</sub>O<sub>2</sub> (30%) were added and evaporated to a small volume. Then, it was cooled, filtered (grade 589/3 blue ribbon), and diluted with 0.5% HNO<sub>3</sub> (in ultra-pure deionized water, 0.05  $\mu$ S cm<sup>-1</sup>) up to the volume of 25 mL. A blank sample was also prepared using a similar experimental procedure (Addis and Abebaw 2017). The purpose of blanks is to trace sources of artificially introduced contamination (Ogink and Klarenbeek 2016).

# Plant sampling and preparation

Savory samples were dried in an electric oven at  $60 \degree C$  until a constant weight was obtained and then powdered. Powdered soil and savory samples were sieved through a 63  $\mu$ m sieve shaker.

Digestion of plant samples was realized according to slightly modified procedure of Mosetlha, Torto, and Wibetoe (2007). Of 1.00 g of each sample was mineralized in an Erlenmeyer flask with 15 mL of conc. HNO<sub>3</sub>, covered with a watch glass and left overnight. After that, the mixture was heated up to 150 °C and H<sub>2</sub>O<sub>2</sub> (30%) was added. Digestion procedure was applied to obtained mixtures to reduce the volume and improve decomposition. Another portion of H<sub>2</sub>O<sub>2</sub> was added and evaporation continued. After cooling, the mixture was filtered (grade 589/3 blue ribbon) and diluted with 0.5% HNO<sub>3</sub> up to 25 mL. A blank sample was prepared in the same way.

# Measurement

All analysis was carried out on iCAP 6000 inductively coupled plasma optical emission spectrometer (Thermo Scientific, Cambridge, UK) that uses an Echelle optical design and a charge injection device solid-state detector. The operating conditions for the ICP-OES instrument were: flush pump rate 100 rpm, analysis pump rate 50 rpm, RF power 1150 W, nebulizer gas flow rate  $0.7 L \text{ min}^{-1}$ , coolant gas flow rate  $12 L \text{ min}^{-1}$ , auxiliary gas flow rate  $0.5 L \text{ min}^{-1}$ , dual (axial/radial) viewed plasma mode and sample uptake delay 30 s.

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All measurements were performed in triplicate. Parameters of conducted ICP-OES analysis based on a calibration curve: wavelength of selected emission lines, correlation coefficient (r), limit of detection (LOD), and limit of quantification (LOQ) of the calibration for each element determination are given in Table 2. The LOD and LOQ values were calculated using the  $3\sigma$  and  $10\sigma$  criterion (Uhrovčík 2014).

The uncertainty of the measurement is affected by many factors such as; incorrect estimation of the amount of analyte in the sample, non-representative sample, ignorance of the factors that influence the measurement process, incorrect values obtained by determining the analyte in the standard reference material (SRM), approximations during the analysis, etc. Figure 1 shows the Ishikawa diagram that describes the influence of various parameters on the method validation process.

# Statistical analysis

Statistical analyses were performed with Statistica version 8 (StatSoft, Tulsa) software packages. All chemical analyses were carried out in triplicate and the results were expressed as mean  $\pm$  standard deviation (SD). The original data set was standardized before statistical analysis to compatible units from a distribution with a mean of 0 and a SD of 1. After standardization, each parameter contributes equally to data set variance and carries equal weight in the principal component (PC) calculation (Dimitrijevic and Miladinovic 2022). At the beginning of the statistical analysis, a correlation matrix was calculated, giving the correlation coefficients between each pair of metals. To determine the statistical significance of variation of accumulation elements in plant and soil during different stages of development, Student's t-test was used. It determines whether any observed differences between the content of elements in plant and soil during different stages of development statistically significant or not. The significance of differences was defined at p < 0.05. The same software carried out hierarchical cluster analysis (HCA) and PC analysis (PCA). Correlation and variability were made at a 95% significance level ( $p \le 0.05$ ).

# Results

# Elements content and variation in savory and its growing soil

The recovery of our analytical method was determined using the SRM (1573a): B, Cr, Mn, Ni, Cu, and Zn. The determined value is reported as value  $\pm$  SD (Table 3).

**Table 2.** ICP-OES parameters: wavelengths ( $\lambda$ ) of the analytical lines, coefficient correlation (r), linearity of the calibration curves, the limit of detection, and quantification (LOD and LOQ) of the calibration curve for each element.

Element	$\lambda$ (nm) – Plant	$\lambda$ (nm) – Soil	r	LOD (mg/kg)	LOQ (mg/kg)
В	208.959	208.959	1.0000	0.0165	0.0548
Si	251.611	251.611	0.9998	0.0408	0.1360
Cr	357.869	357.869	0.9999	0.0476	0.1587
Mn	257.610	257.610	0.9999	0.1255	0.4182
Ni	341.476	231.604	1.00000	0.1594	0.5313
Cu	327.396	327.396	0.99997	0.0190	0.0632
Zn	213.856	213.856	0.99984	0.0031	0.0104



Figure 1. Ishikawa diagram.

**Table 3.** The determined values (mg kg<sup>-1</sup>), certified values (mg kg<sup>-1</sup>), and RSD (%) for the quantification of B, Cr, Mn, Ni, Cu, and Zn in standard reference material (SRM 1573a).

Element	Certified value (mg $kg^{-1}$ )	Determined value (mg $kg^{-1}$ )	Recovery (%)	RSD (%)
В	33.13 ± 0.42	32.1 ± 0.39	96.9	1.2
Cr	$1.988 \pm 0.034$	$2.04 \pm 0.03$	102.6	1.5
Mn	246.3 ± 7.1	241.0±6	98.7	2.5
Ni	$1.582 \pm 0.041$	$1.59 \pm 0.02$	100.5	1.3
Cu	$4.70 \pm 0.14$	$4.63 \pm 0.04$	98.5	0.9
Zn	$30.94 \pm 0.55$	$31.2 \pm 0.4$	100.8	1.3

Recovery was expressed as percentage differences between the measured concentration and the certified value to SRM (%). Method precision was evaluated as repeatability and is expressed through the relative SD as a percentage (%). Differences between certified values and quantified concentrations were below 10%. The recovery values were in range of 93.6% and 106.2%. All the results are presented in Table 3.

TE contents in savory during different stages of development and its growing soil were given in Table 4. The total level of elements in the soil reflects the geological and climatic origin of the soil. In this research contents of selected elements were within the specified soil values (Sparks 2003). The highest concentration values (mg kg<sup>-1</sup>) were recorded for Mn: 135–221, while the lowest values were noted for boron: 6.8–20.7. It is interesting to point out that the lowest contents of the studied elements in the soil, except for silicon, were recorded in June (vegetative stage).

The coefficients of variation (CVs) are shown in Figure 2. CVs of the soil samples decreased in the sequence Si > Cr > B > Zn > Ni > Mn > Cu, while plant samples followed the order Cr > Ni > Si > Zn > Cu > Mn > B.

# Soil-to-plant transfer factor

Soil-to-plant TF indicates the uptake and accumulation behavior of elements in *S. kitai-belii* (Table 5). TF was calculated as the concentration of TE in plant over that in soil

	Elements								
Months	В	Si	Cr	Mn	Ni	Cu	Zn		
			Plant						
M1	14.1 ± 0.8 <sup>b</sup>	$3.9 \pm 0.2^{\circ}$	$0.17 \pm 0.01^{\circ}$	$5.8 \pm 0.4^{b}$	$2.7 \pm 0.1^{a}$	$2.3 \pm 0.1^{\circ}$	$7.9 \pm 0.7^{e}$		
M2	$14.3 \pm 0.8^{b}$	1.79 ± 0.09 <sup>e</sup>	$0.18 \pm 0.01^{\circ}$	$4.1 \pm 0.3^{d}$	$0.66 \pm 0.05^{\circ}$	$2.1 \pm 0.1^{d}$	$8.2 \pm 0.6^{d}$		
M3	$10.1 \pm 0.7^{e}$	$5.6 \pm 0.4^{b}$	$1.2 \pm 0.1^{a}$	$3.8 \pm 0.3^{d}$	$0.62 \pm 0.05^{\circ}$	$2.81 \pm 0.2^{b}$	$20.1 \pm 0.9^{a}$		
M4	$13.8 \pm 0.8^{\circ}$	$6.7 \pm 0.5^{a}$	$0.25 \pm 0.02^{b}$	$5.7 \pm 0.4^{b}$	$0.59 \pm 0.05^{\circ}$	$2.4 \pm 0.1^{\circ}$	10.9 ± 0.7 <sup>c</sup>		
M5	$14.9 \pm 0.8^{a}$	$2.5 \pm 0.2^{d}$	$0.22 \pm 0.02^{bc}$	$6.7 \pm 0.5^{a}$	1.3 ± 0.1 <sup>b</sup>	$3.0 \pm 0.1^{a}$	$13.8 \pm 0.8^{b}$		
M6	10.5 ± 0.7 <sup>d</sup>	$4.9 \pm 0.4^{b}$	$0.32 \pm 0.03^{b}$	$4.5 \pm 0.3^{\circ}$	0.55 ± 0.01 <sup>c</sup>	$0.70 \pm 0.06^{e}$	$8.2 \pm 0.7^{d}$		
			9	Soil					
M1	$6.8 \pm 0.5^{f}$	$195 \pm 12^{a}$	$4.4 \pm 0.4^{e}$	135 ± 10 <sup>c</sup>	5.7 ± 0.5 <sup>d</sup>	$7.2 \pm 0.6^{e}$	$8.5 \pm 0.7^{e}$		
M2	$20.7 \pm 0.9^{a}$	126 ± 11 <sup>c</sup>	$20.9 \pm 0.9^{a}$	164 ± 12 <sup>b</sup>	$16.3 \pm 0.8^{a}$	10.2 ± 0.7 <sup>b</sup>	28 ± 1 <sup>c</sup>		
M3	$9.2 \pm 0.7^{e}$	24 ± 1 <sup>e</sup>	$14.4 \pm 0.8^{\circ}$	$212 \pm 14^{a}$	10.9 ± 0.7 <sup>c</sup>	9.2 ± 0.7 <sup>d</sup>	$30 \pm 2^{b}$		
M4	16.3 ± 0.8 <sup>b</sup>	112 ± 10 <sup>d</sup>	$16.9 \pm 0.8^{b}$	172 ± 12 <sup>b</sup>	12.5 ± 0.7 <sup>b</sup>	10.3 ± 0.7 <sup>b</sup>	$32 \pm 1^{a}$		
M5	$15.0 \pm 0.7^{\circ}$	112 ± 11 <sup>d</sup>	12.2 ± 0.7 <sup>d</sup>	175 ± 13 <sup>b</sup>	11.1 ± 0.7 <sup>c</sup>	9.5 ± 0.7 <sup>c</sup>	25 ± 1 <sup>d</sup>		
M6	$12.6 \pm 0.7^{d}$	140±11 <sup>b</sup>	$16.7 \pm 0.8^{b}$	$221 \pm 14^{a}$	$13.2 \pm 0.7^{b}$	$11.8 \pm 0.7^{a}$	$30 \pm 2^{b}$		

Values are the mean  $\pm$  standard deviation (n = 3).

Values with different letters within columns are statistically different at p < 0.05 by paired Student t test.



Figure 2. Elements coefficients of variation in S. kitaibelii and soil samples.

Table 5. S. kitaibelii soil-to-plant transfer factors.

Months	Elements								
	В	Si	Cr	Mn	Ni	Cu	Zn		
M1	2.05	0.02	0.04	0.04	0.48	0.33	0.93		
M2	0.69	0.01	0.01	0.03	0.04	0.20	0.30		
M3	1.10	0.23	0.08	0.02	0.06	0.31	0.67		
M4	0.85	0.06	0.01	0.03	0.05	0.24	0.34		
M5	1.00	0.02	0.02	0.04	0.11	0.32	0.56		
M6	0.83	0.04	0.02	0.02	0.04	0.06	0.27		

(TF = [TE plant]/[TE soil]). A plant could be considered to be an accumulator of the studied element when TF > 1 (Márquez-García and Córdoba 2010).

The highest TF values were recorded for five elements (except Si and Cr) in the vegetative stage. TF values of boron in June and August were greater than 1 and higher than those of other elements, during the studied stages of development. Therefore, savory can be considered an accumulator of this element. In addition, with a TF value of 0.93 in vegetative stage, *S. kitaibelii* showed a good tendency to accumulate Zn.

## Correlation analysis between soil and plant elements

Correlation analysis (CA) between the content of elements in the plant and soil samples was conducted to investigate their interaction. The results are shown in Table 6. As can be seen, eight strong negative correlations (Ratner 2009) were identified.

# Hierarchical cluster analysis

The TEs of savory and soils in three different stages of development were subjected to chemometrics analysis to detect any interactions between them. The similarity of the different stages of savory and similarity between analyzed elements was assessed using HCA. HCA is a multivariate technique to classify objects of a system into categories or clusters based on their similarities (Johnson and Wichern 2002). The distance between the two objects indicates their similarity, i.e., dissimilarity.

HCA was performed by Ward's method using Pearson's correlation as a measure of similarity. When two objects are close, it indicates a significant similarity. The distance will be less and get closer to 0 as the correlation goes to 1. The distance was reported as  $D_{link}/D_{max}$ , representing the quotient between the linkage distances for a particular case divided by the maximal linkage distance (Singh et al. 2004).  $D_{link}$  is the distance between the variables that are grouped and  $D_{max}$  is the maximum distance between the variables. The results are shown as a dendrogram in Figures 3 and 4.

# Principal component analysis

The PCA method uses and presents more information, unlike HCA (Patras et al. 2011). The goal of the HCA is to partition the samples into homogeneous groups-clusters, such that the within clusters similarities are large compared to the between-clusters similarities. On the other hand, PCA aims to reduce and extract original variables in a smaller number of underlying variables, to reveal the interrelationships between the variables. Also, to find the optimum number of extracted PCs.

Analyzed elements were correlated with two PCs with 74.79% of the total variance. This is an acceptably large percentage. The results are shown in Figure 5. The first PC describes the maximum possible variation that can be projected onto one dimension; the second PC

	B-S	Si-S	Cr-S	Mn-S	Ni-S	Cu-S	Zn-S		
B-P	0.43	0.51	-0.18	-0.84	-0.07	-0.34	-0.39		
Si-P	-0.39	-0.30	-0.04	0.38	-0.22	0.12	0.32		
Cr-P	-0.38	<b>-0.87</b>	0.09	0.63	-0.02	-0.02	0.38		
Mn-P	-0.05	0.46	-0.53	-0.53	-0.46	-0.32	-0.41		
Ni-P	-0.53	0.58	<b>-0.89</b>	-0.81	-0.83	-0.91	-0.96		
Cu-P	-0.05	-0.39	-0.29	-0.39	-0.30	-0.61	-0.16		
Zn-P	-0.27	-0.90	0.00	0.46	-0.08	-0.13	0.35		

Table 6. Correlation analysis between plant and soil trace elements.

Bold values indicate strong negative correlations. P: plant; S: soil



Figure 3. Dendrogram of different stages of development of S. kitaibelii.



Figure 4. Dendrogram of element contents in *S. kitaibelii* and soil during different stages of development.

captures the second most and so on (Anderson et al. 1999). In this case, the first component explained 47.20% while the second component explained 27.59% of the total variance. Ni accumulation in plants and Zn content in soil are the most important contributors to the formation of PC1, 14.3% and 13.3%, respectively (Table 7). At the same time, the highest contribution on PC2 had Cr-P (17.5%) and B-S (16.2%).



Figure 5. Principal component analysis for elements and different stages of development of *S. kitaibelii.* 

Variable	PC1	PC2
B-P	6.1	7.9
Cr-P	4.2	17.5
Cu-P	1.5	4.2
Mn-P	6.8	0.7
Ni-P	14.3	0.9
Si-P	1.2	6.6
Zn-P	1.9	15.9
B-S	2.4	16.2
Cr-S	10.7	5.8
Cu-S	9.9	5.5
Mn-S	11.1	2.1
Ni-S	8.9	8.8
Si-S	7.4	7.6
Zn-S	13.3	0.3

Table 7. Contribution of variables to the formation of PC1 and PC2 (%).

Table 8. Contribution of elements in S. kitaibelii to RDI (%).

	Elements							
Months	В	Si	Cr	Mn	Ni	Cu	Zn	
RDI (mg day)	1	33.5	0.025	1.8	0.05	0.9	11	
M1	21.2	0.2	10.2	4.8	81.0	3.8	1.1	
M2	21.5	0.1	10.8	3.4	19.8	3.5	1.1	
M3	15.2	0.3	72.0	3.2	18.6	4.7	2.7	
M4	20.7	0.3	15.0	4.8	17.7	4.0	1.5	
M5	22.4	0.1	13.2	5.6	39.0	5.0	1.9	
M6	15.8	0.2	19.2	3.8	16.5	1.2	1.1	

RDI: recommended dietary intake

# Contribution of elements in S. kitaibelii to recommended dietary intake

We calculated the contribution of all elements to recommended dietary intake (RDI) (US National Academies, Institute of Medicine 2001). The results are presented in Table 8.

The contribution of the elements was calculated on the assumption of consuming three cups of tea, i.e.,  $3 \times 5$  g of dried plant per day. As can be seen, the percentage of potentially possible daily intake of the elements is in the range of 0.1% (Si) to 81% (Ni).

# Discussion

# **Elements content variation**

The content of TEs in the plant depends on several factors such as plant species, factors of soil, stage of maturity, and seasonal and temperature effects (Kabata-Pendias 2011). In savory from Serbia sufficient contents of studied elements were determined, except chromium (Watanabe et al. 2007). In the plant, the boron content is the highest:  $10.5-14.9 \text{ mg kg}^{-1}$  and the chromium content is the lowest:  $0.17-1.2 \text{ mg kg}^{-1}$ . In an investigation related to content of macroelements and TEs in two species of *Satureja* genus, a very similar content of elements was established, with a note that the content of chromium was slightly higher (Dunkić et al. 2012).

If we accept the criterion that the CV is acceptable in the range of 20-30% (Pimentel-Gomes 2009), it can be concluded that Mn concentrations in the soil and in the plant were stable, and the CVs were <23%. Concentrations of Cu in savory were unstable, and the CV was large. By contrast, its concentrations in the soil were very stable, and the CV was small. Therefore, the absorption of this element might be affected by the climatic factors in the growth location of savory. On the other hand, concentration of boron is the most stable in the plant, compared to all other elements, CV was 16.4%. By contrast, its concentrations were not stable in the soil samples, and the CV was large. So, it can be assumed that the absorption of B is related to biological characteristics of *S. kitaibelii*, rather than its soil conditions (Filip and Tack 2010). It is an interesting fact that the three elements are not classified as essential in plants, unlike the other four examined elements.

# Relationships between elements content in savory and its growing soil

Correlations between samples were determined based on the Pearson coefficient; higher than 0.7 can be considered a significant correlation and a coefficient having a value above 0.9 indicates a strong correlation between variables (Ratner 2009). The results of the CA indicate that silicon, Mn, and Cu in the plant samples were not related to any element in the soils. Also, boron in the soils was not related to any element in the plant samples. Obviously, that the total particular element content in soil negatively affects the element content in plant samples. It can be assumed that the contents of the elements in the savory are affected by available forms of TEs in the soil (Kabata-Pendias 2011).

# Chemometric analysis

For a quality comparison, hierarchical cluster analysis was applied to group months based on the accumulation of elements in savory and its growing soil. HCA yields a dendrogram (Figure 3), suggesting two statistically significant clusters at  $(D_{link}/D_{max}) \times 100 < 50$ , cluster A and cluster B. Cluster A is divided into two sub-clusters (A' and A"). The strongest clustering is observed for September (M4) and November (M6), with minimal distance, which showed close association with October (M5). This observation indicates a significant similarity among these samples. In addition to the existence of subclusters in this cluster, M2 (July) separation from other months was observed. Mentioned samples belong to subcluster A'. The second subcluster (A") was composed of June (M1). The most significant distance between the samples was recorded in clusters A and B, indicating the differences. Cluster B was constituted by August (M3).

Figure 4 shows a dendrogram of cluster analysis for the mean of element contents in plant and soil in different stages of development. The main objective of HCA is to investigate similarities between the accumulated elements and indicate the reason for their clustering. Division to three clusters, A, B, and C (condition: (D<sub>link</sub>/D<sub>max</sub>)  $\times 100 < 50$ ) indicates a different content in the soil and accumulation of elements by the plant. Elements content in soil (Mn-S, Zn-S, Cu-S, Ni-S, Cr-S, and B-S) are associated in cluster A. This cluster is divided into two subclusters. Cr-S, Ni-S, and B-S form a separate subcluster (A'), while Cu-S, Zn-S, and Mn-S constitute the second subcluster (A"). The smallest distance was recorded between Cr-S and Ni-S, indicating the significant correlation between these two elements in the soil. Nickel concentrations are frequently associated with high concentrations of iron, zinc, and chromium in soil (Barker and Pilbeam 2007). Cluster B contains only elements accumulated in the plant, such as Cr-P, Zn-P, Cu-P, and Si-P. The strongest clustering within this cluster is between Cr and Zn. Within the cluster C, there are two subclusters (C' and C"), one of which is important to point out because it indicates a correlation between Mn and boron in the plant. The second sub-cluster was composed of Ni-P and Si-S.

The PCA results are in accordance with the HCA analysis but using this method, we raised our study to the next level to display the connection between the accumulation of elements and different stages of plant development. Not all of the PCs are of equal importance, only the first few of them contain the important data. The number of significant components can be determined using the criteria established by Kaiser (Kaiser 1960) and, according to him, the important PCs are those whose value is higher than one. The PCA pointed out M2, M4, and M6 on the plot's left side, suggesting that Zn-S, Cu-S, Cr-S, Ni-S, and B-S, which were found in the same quadrant, are dominant elements in the soil in these months. This grouping corresponds to cluster A (Figure 4). As illustrated in Figure 5, the vectors of the variables Cu-S and Cr-S are parallel, indicating a strong correlation (Ratner 2009).

The vectors of variables Cr-P and Zn-P occupy an acute angle, indicating a significant correlation. These elements and Si-P and Mn-S are co-located in the higher left-hand quadrant of Figure 5, together with M3, suggesting that they have a high content in this stage of development. Ni-P is co-located in the higher right-hand quadrant, in the immediate vicinity with M1, suggesting high content of Ni-P in this month. M5 (October) occupied a location in the fourth quadrant of the figure. In this month we have the highest concentration of Mn in the plant, which is also located in this area.

# Potential aspects of health promotion

As we have said, TEs are helpful for proper growth, development, preservation, and recovery of organism health. They are important components of enzymes that donate or accept electrons, regulating important biological processes through actions such as assisting the binding of molecules to receptor sites on cell membranes. Additionally, some TEs provide structural stability to important biological molecules (Anal and Chase 2016).

Boron is not an essential TE in human health, but its biochemical function is very important in numerous biological functions, including calcium metabolism, growth, and maintenance of bone tissue. Also, boron reduces the risk of certain types of cancer, the development of arthritis, and associated heart disease symptoms. Further, it accelerates wound healing, reduces pain in gynecological diseases, and kidney stones by reducing cytokines (Rondanelli et al. 2020). Our research has determined that *S. kitaibelii* is the accumulator of this element. The conclusion is that a significant percentage of daily intake of B, Cr, and Ni, can be provided with three cups of tea per day of plants collected in the vegetative and flowering stages. It should be said that the greatest contribution of boron to RDI was recorded in October (22.4%). The value was insignificantly lower in July (21.5%). Considering that it is better, i.e., healthier, to use a younger plant, we recommend for savory collection the flowering stage.

# Conclusions

In this study, combined chemical and chemometric analysis of seasonal variation in TE contents, in the *S. kitaibelii* Wierzb. ex Heuff. and its growing soil, over the course of six months was done, with emphasis on potential aspects of health promotion. On the base statistical and chemometric analysis, it can be concluded that there is a variation in TE content of studied soil and plant samples, caused by seasonal variation. The lowest contents of the studied elements in the soil, except for silicon, were recorded in June (vegetative stage). In this stage of development, the highest values of soil-to-plant TF were recorded for five elements (except Si and Cr). Thanks to this research, it has been established that *S. kitaibelii* from Serbia can be considered a potential source of valuable TEs and an accumulator of boron. A significant percentage of daily intake of B (15.2–22.4%), Cr (10.2–72.0%), and Ni (16.5–81.0%), can be provided with three cups of tea per day of plants collected in the vegetative and flowering stages.

# **Disclosure statement**

The authors have declared that no competing interests exist.

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# ORCID

Dragoljub Miladinović D http://orcid.org/0000-0003-4849-1067 Marija Dimitrijević D http://orcid.org/0000-0003-1816-0400

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